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Preparation of alpine ski pistes



Preparation of alpine ski pistes



This Power Point Presentation is intended for FIS alpine technical delegated persons. It has been written by Mathieu Fauve and Hansueli Rhyner from the Swiss Federal Institute for Snow and Avalanche Research, Davos and completed by Thomas Gurzeler. It is a tool for the alpine race supervisors, officials, coaches and Athletes, who strive for a safe race. The use and modifications of this presentation are possible only with permission from Hansueli Rhyner.

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Preparation of alpine ski pistes



Aim

The aim is to transform a soft snow into a:

- hard
- homogeneous

piste

Proceed

Use of **mechanical equipment** taking into account the **physical properties of snow** and the **meteorology**

Special

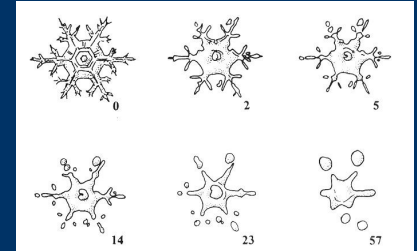
Race pistes have to be prepared with **special methods** in order to obtain very high strength



Preparation of alpine skiing slopes



physical properties of snow



snow and meteorology



mechanical handling of snow



preparation and maintenance of race
pistes



Physical properties of snow



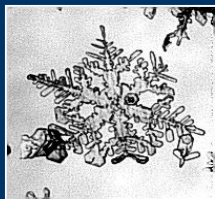
Snow is a very particular material:

- ➔ composed of air and water in all its forms: solid (ice), gas (water vapour) and possibly liquid (liquid water)
- ➔ near its melting point (0°C), so extremely sensitive to variations of temperature and pressure and reacts rapidly
- ➔ exists in different forms and is subjected to a continuous transformation

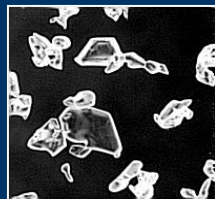
Snow grains



new snow



faceted crystals



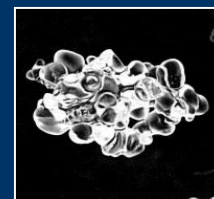
fragmented precipitation particles



surface hoar



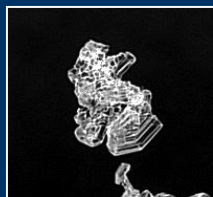
melt-freeze snow



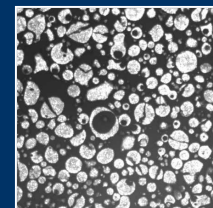
rounded grains



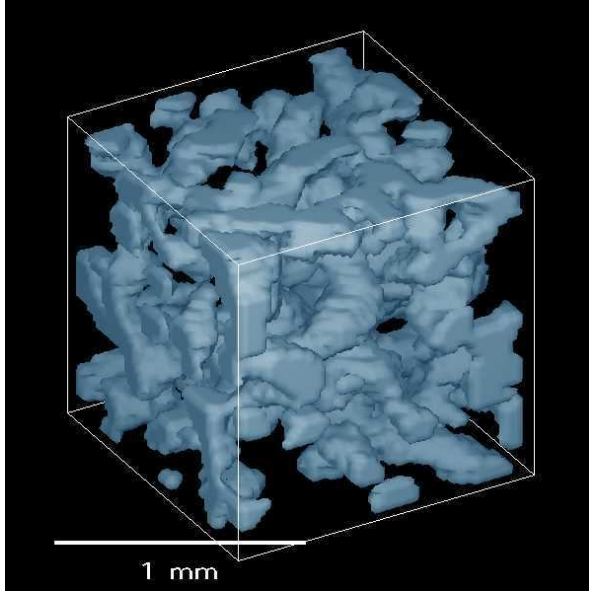
depth hoar



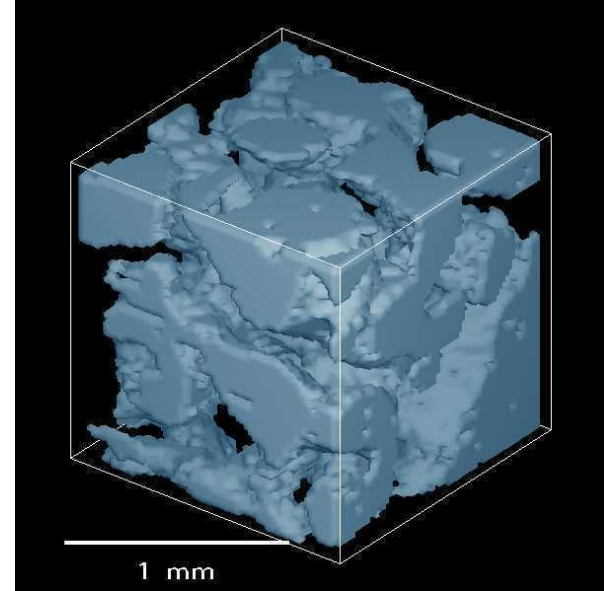
machine-made snow



Snow density



low density snow (220 kg/m³)



high density snow (516 kg/m³)

- new snow density: between 50 and 250 kg/m³
- average ski slope density : 480 kg/m³
- downhill racing slope density: 300 - 500 kg/m³
- super-G racing slope density: approx. 550 kg/m³
- slalom racing slope density: approx. 600 kg/m³

Snow strength



snow

- ➡ grain type
- ➡ density
- ➡ bonds between the grains:
strength



dry snow:
stellar interlocking
sintering

wet snow:
capillary
freezing

Strength by stellar interlocking



- concerns only new snow stellar crystals
- branches are connected to each other

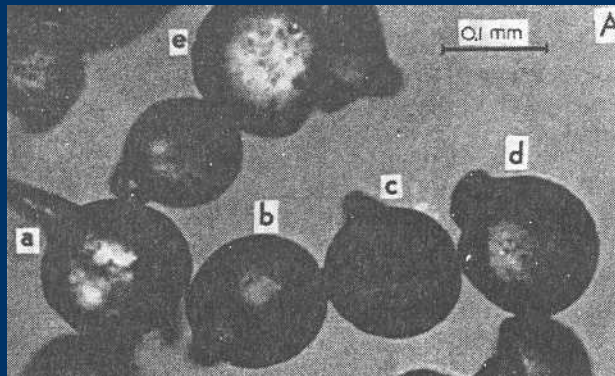


➔ **weak and non-lasting strength**

Strength by sintering



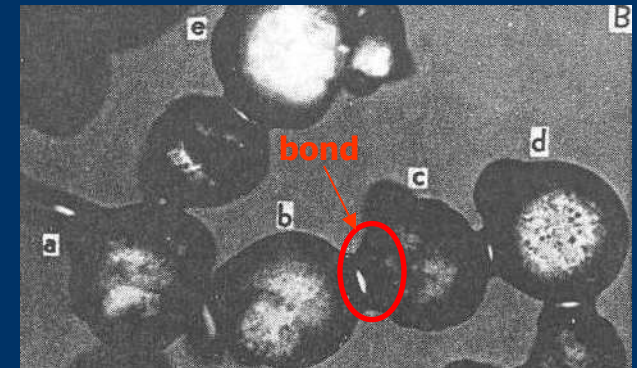
sintering: building of solid bonds between snow grains by water vapour transport



$T = -1.5^{\circ}\text{C}$



after 165 min



(Kuroiwa, 1974)

➡ the building of the bonds and the increasing of their size depend on the following parameters:

- **temperature** (process faster near 0°C)
- **grain type** (shape, mean size, size distribution)
- **density** (number of contact points)
- **time** (process needs time)

Strength of wet snow



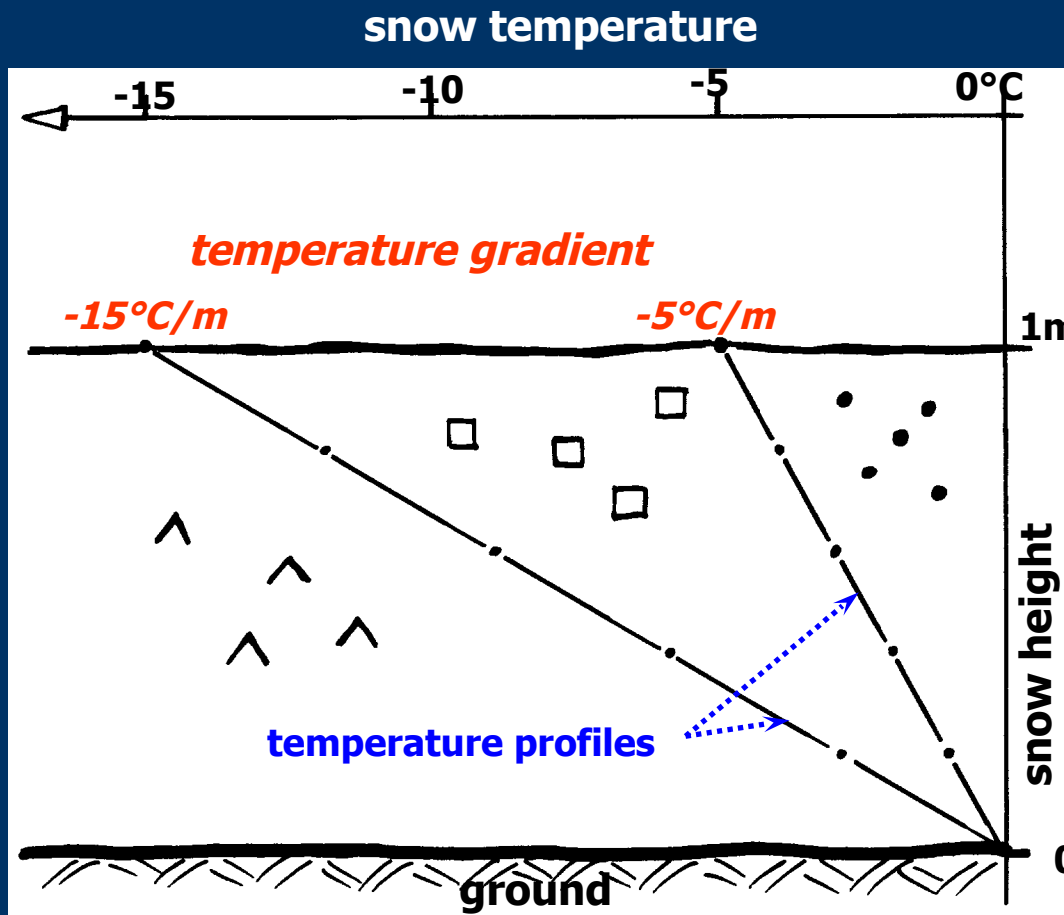
- liquid water is held on grains by capillarity as long as its volume is lower than 5 to 10% (depending on the grain size) of the total volume. Small grains can hold more water than bigger ones.
- when the liquid water content (L.W.C.) is high, bonds melt and snow becomes softer
- when the liquid water freezes, strong bonds built between the grains



Snow metamorphism



Snow metamorphism depends on the temperature gradient in the snowpack

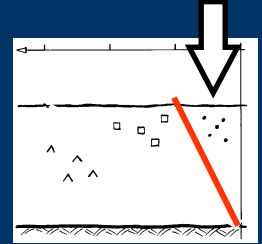
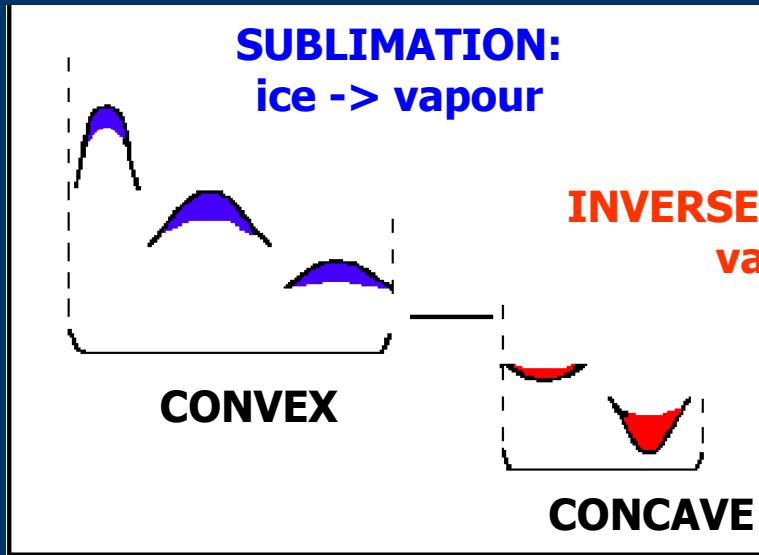


$$T_G = \frac{dT}{dz}$$

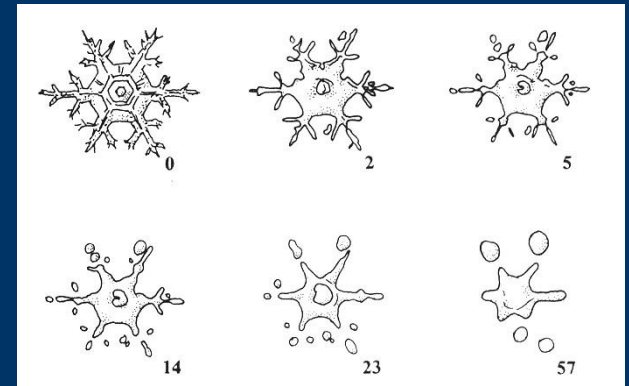
Destructive metamorphism



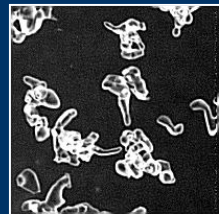
water vapour pressure



$$T_G < 5 \text{ } ^\circ\text{C/m}$$



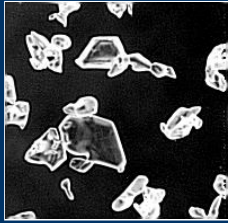
- mass transport from convex to concave zones
- building of small (0.2 to 0.75 mm) round grains
- formation and size increase of inter-granular bonds (sintering)
- slow process



Constructive metamorphism



medium gradient

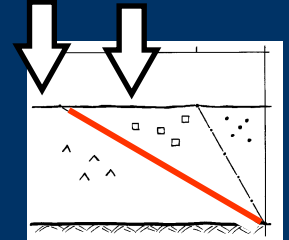


**faceted grains
(1 to 3 mm)**

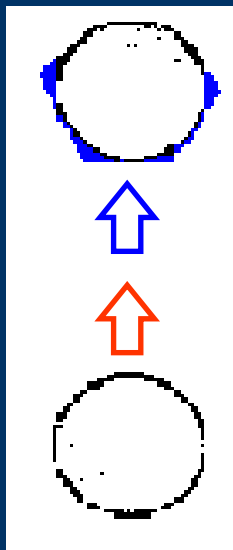
high gradient



**depth hoar
(1.5 to 5 mm)**



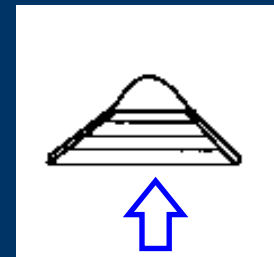
$T_G > 5 \text{ } ^\circ\text{C/m}$



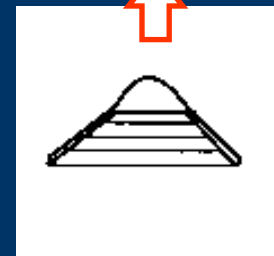
condensation



sublimation

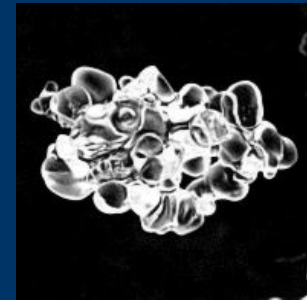
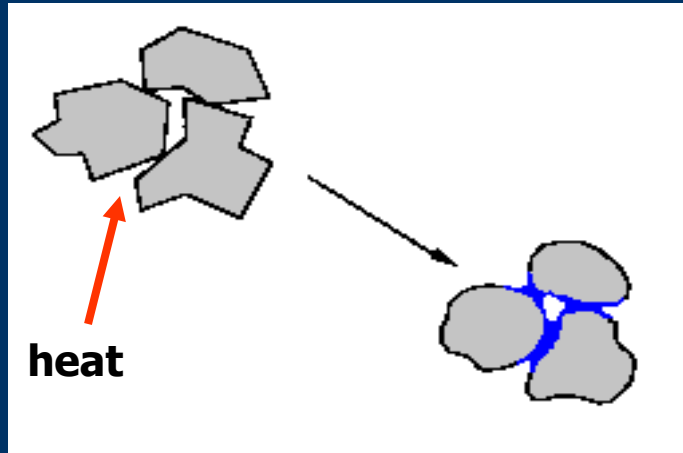


cold

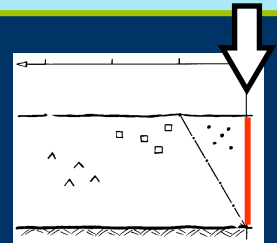


warm

Melt-freeze metamorphism



0.5 to 4 mm



T = 0°C
L.W.C. > 0

- grains become rounder and bigger
- influence of the liquid water content:
 - non-saturated snow (L.W.C. < 8 to 15% vol.)
 - ➔ clusters
 - saturated snow (L.W.C. > 8 to 15% vol.)
 - ➔ no cohesion

Mechanical properties of snow



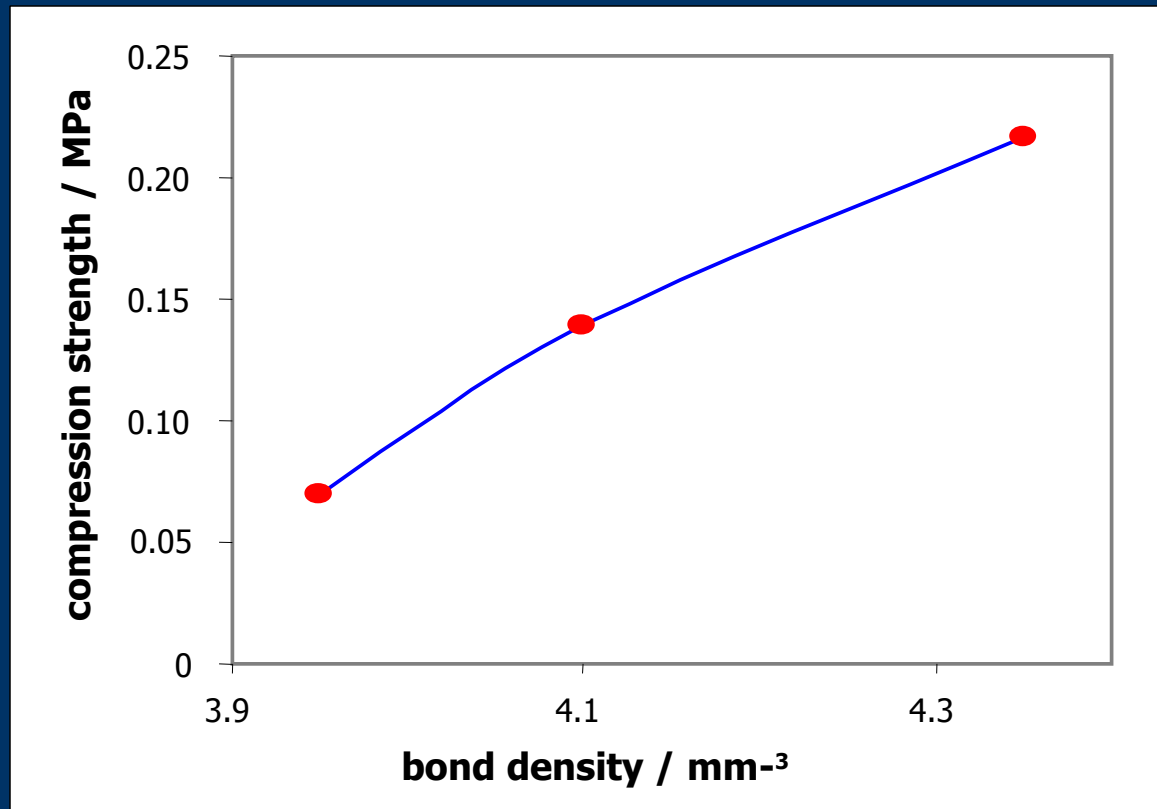
The mechanical properties of snow mostly depend on the following parameters:

- bonds
- density
- temperature
- liquid water content

Mechanical properties of snow



bonds

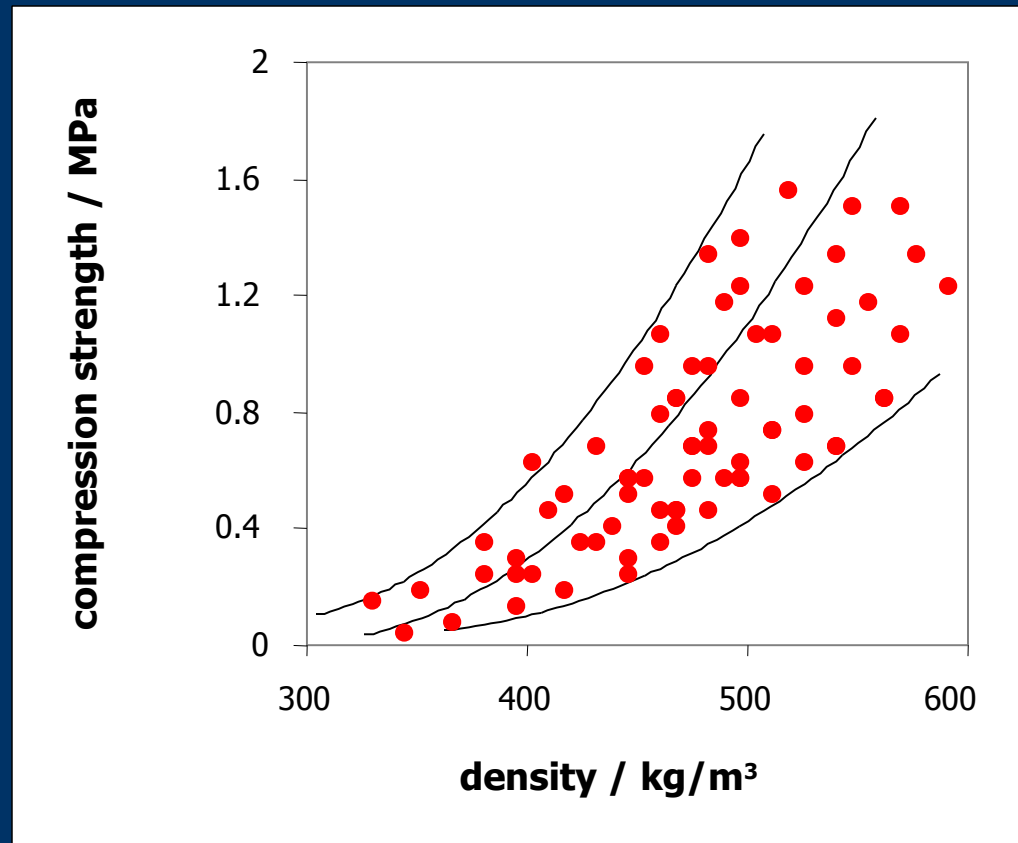


➡ the more bonds that exist and the larger they are, the higher the snow's resistance is

Mechanical properties of snow



density

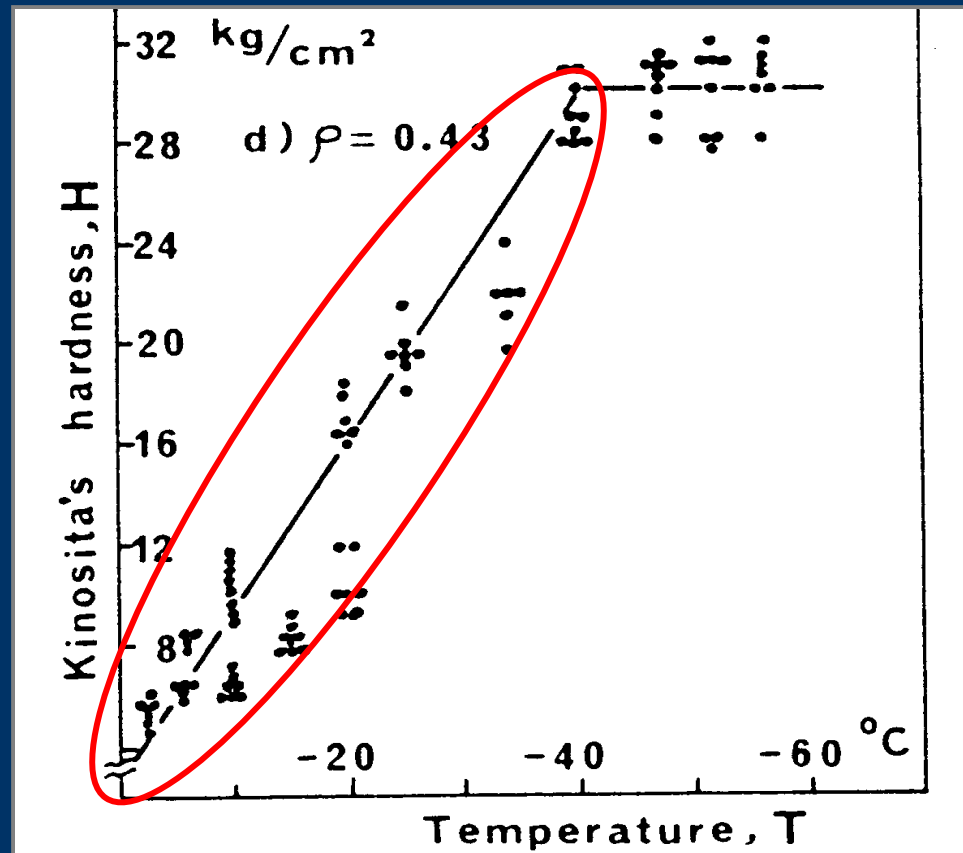


➡ usually, the denser the snow, the more resistant and tough it is

Mechanical properties of snow



temperature



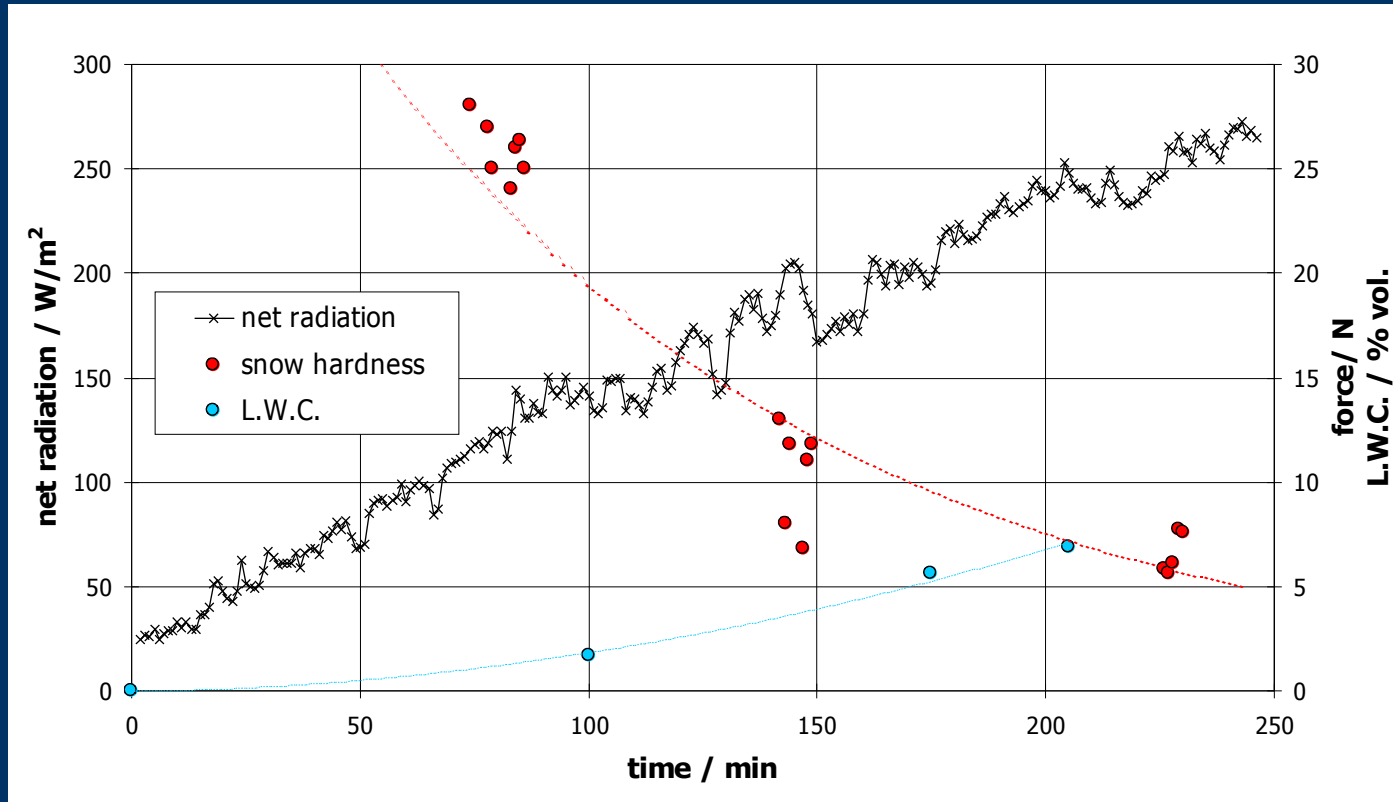
(Tusima, 1974)



➡ the colder dry snow is, the more resistant and tough it becomes

Mechanical properties of snow

liquid water content ($T_{\text{snow}}=0^{\circ}\text{C}$)



➡ for a snow at 0°C and high solar incoming radiation, the liquid water content at the snow surface increases rapidly and snow becomes softer

machine-made snow

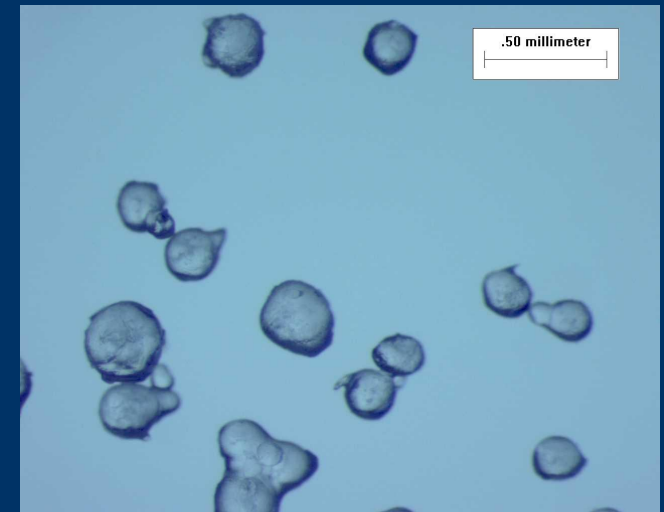
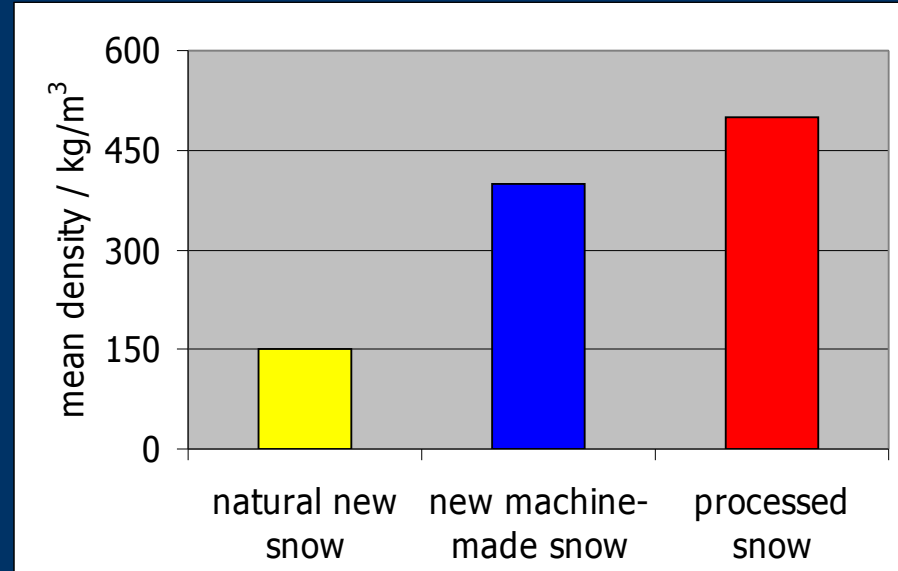


characteristics

- high density:
300 to 500 kg/m³
- small and round grains
(0.1 to 0.9 mm)



- ➡ resistant snow
- ➡ needs little compaction

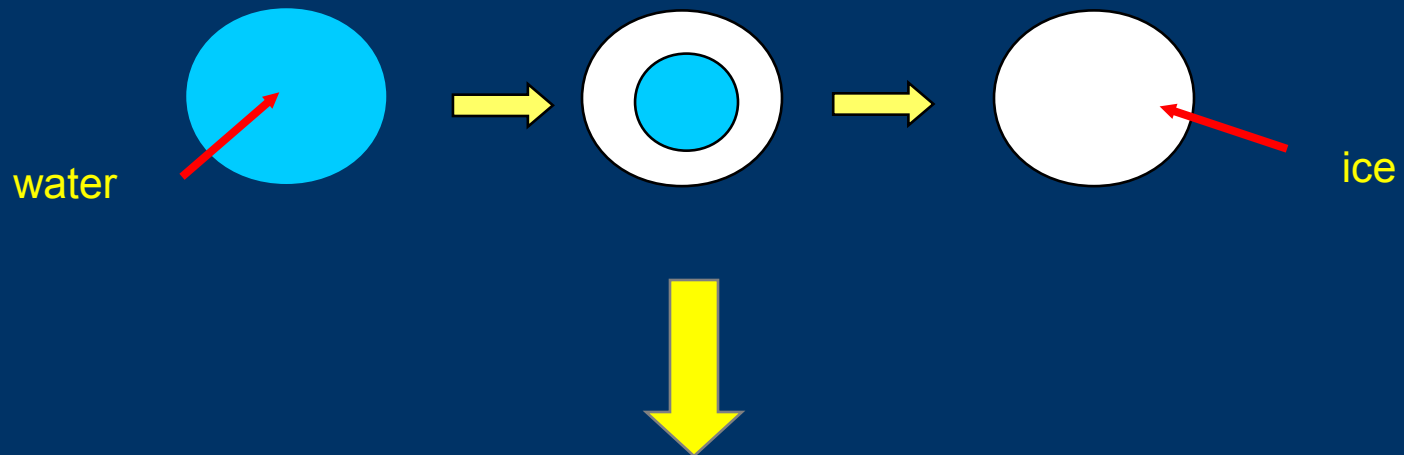


machine-made snow



characteristics

- Warning: risks of incomplete freezing of water droplets



- a curing time (for complete freezing) is needed before grooming the snow

Snow and meteorology



The different properties of snow depend mostly on one parameter: its temperature.

Snow temperature depends on heat exchanges between snow and air: heat balance

The heat balance at the snow surface = difference between gain and loss of heat energy



balance is positive

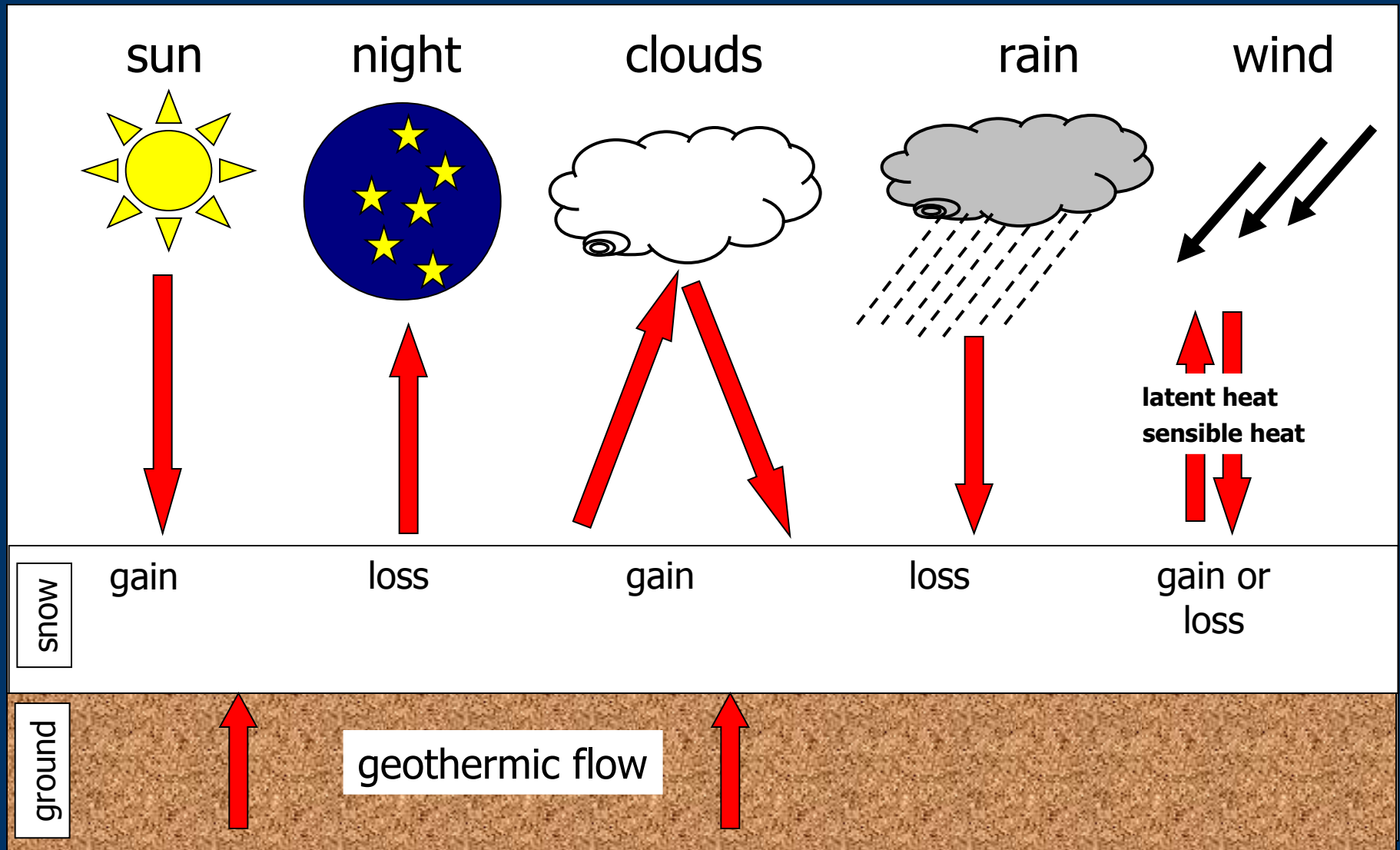
snow temperature increases
at 0°C: snow starts melting

balance is negative

snow temperature
decreases



Heat balance

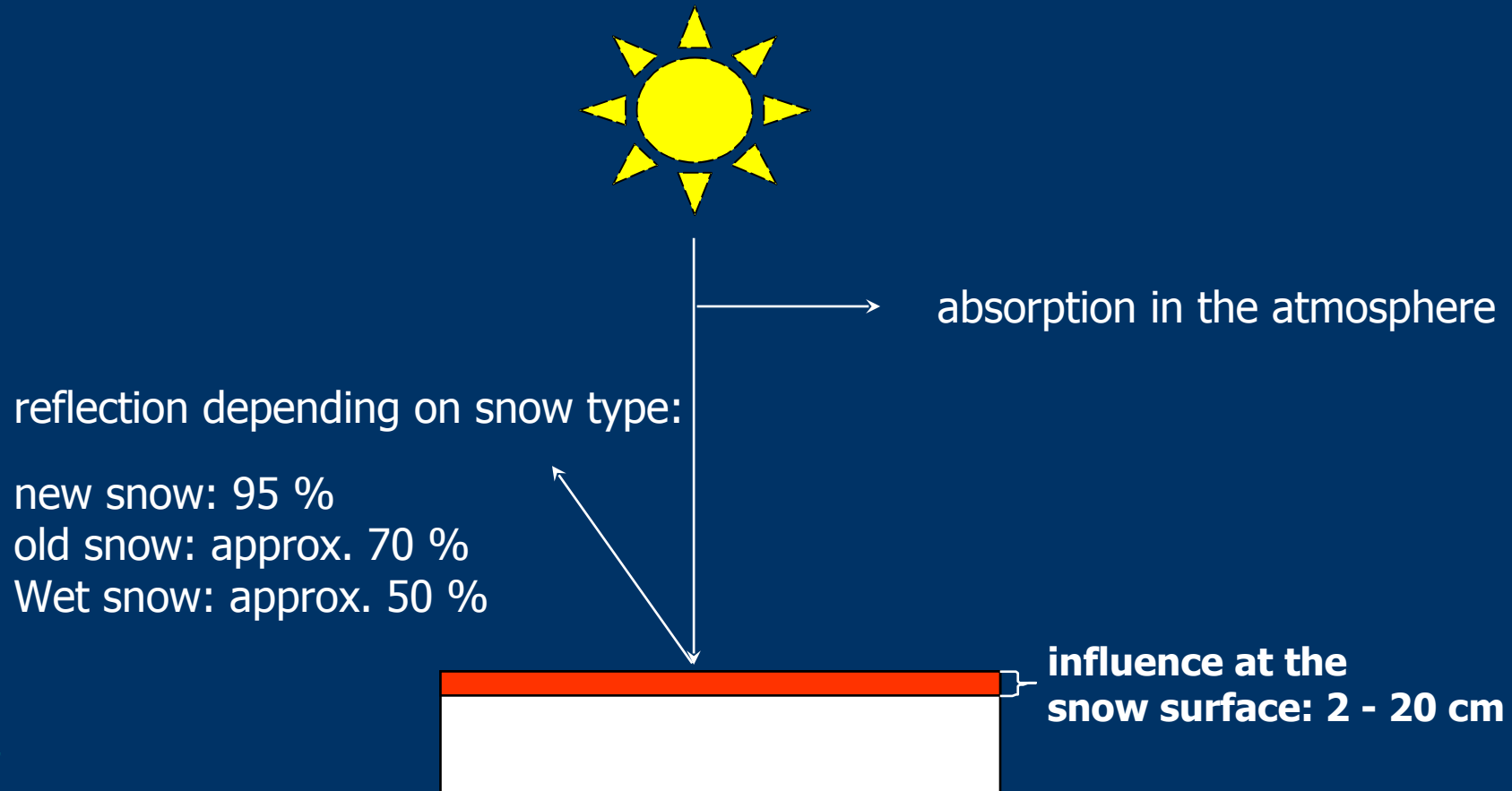


Gain and loss of heat energy at the snow surface

Incoming solar radiation at the earth's surface



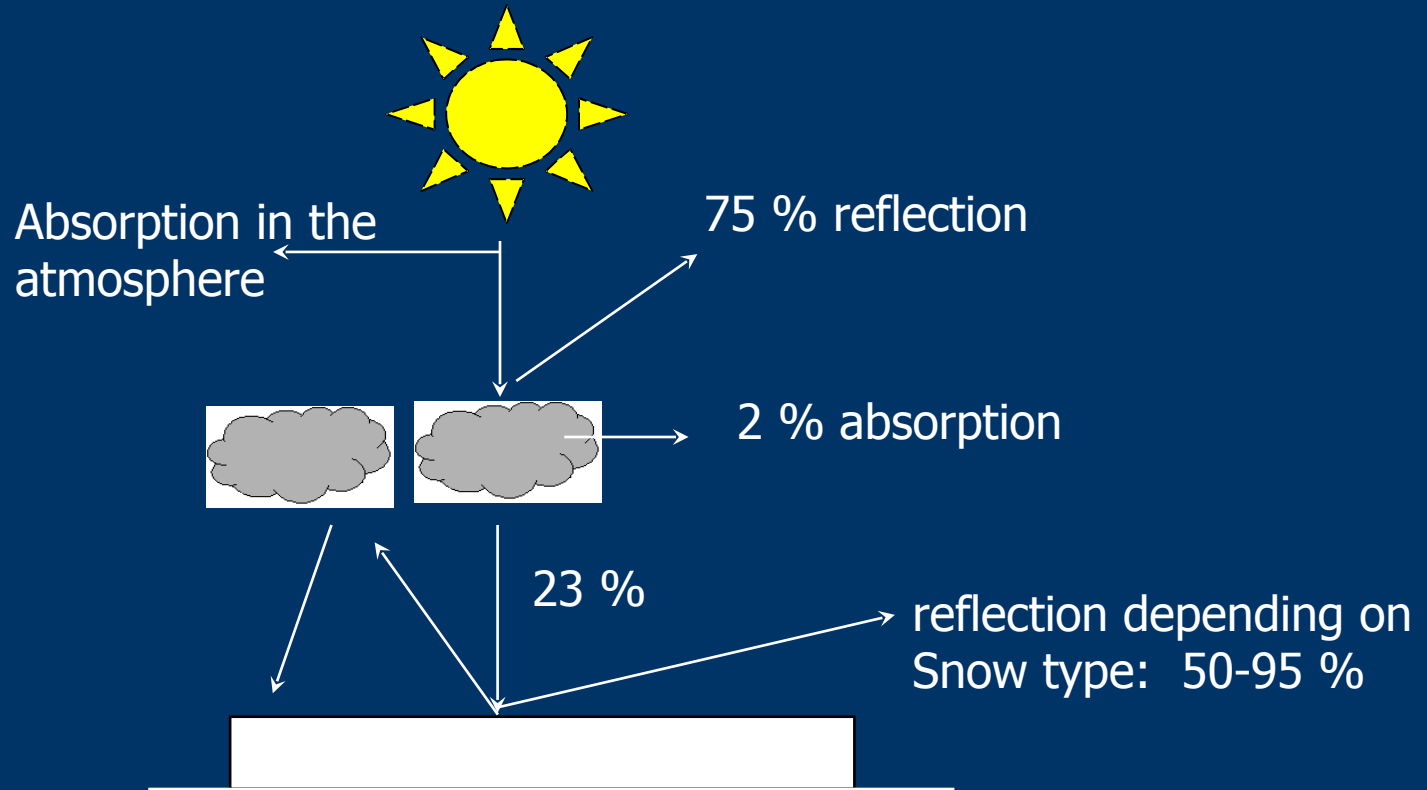
incoming solar radiation at the snow surface on a sunny day



Incoming solar radiation at the earth's surface



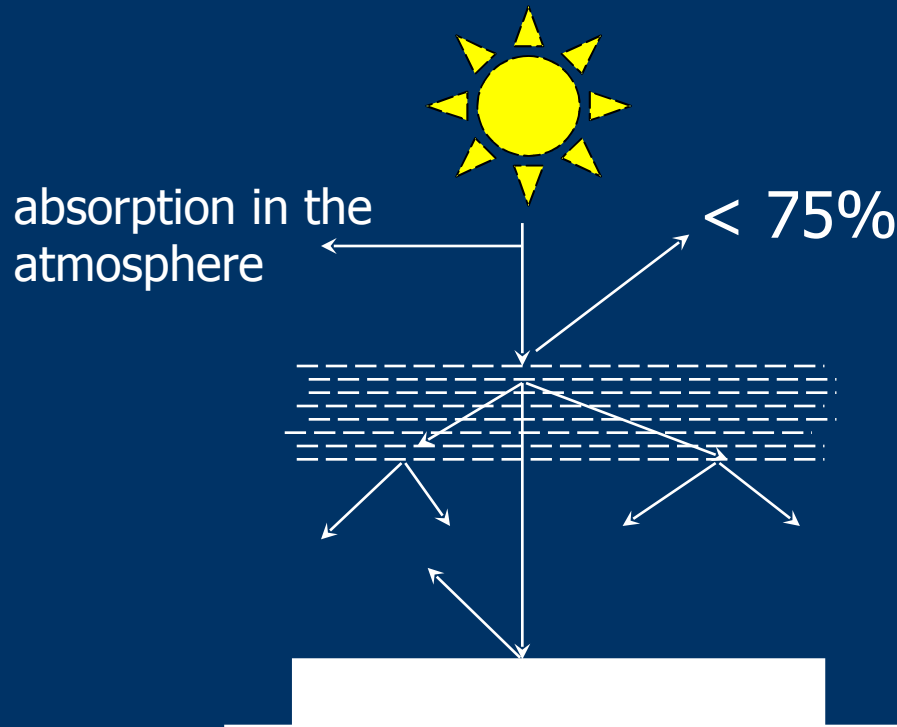
incoming solar radiation at the snow surface on a cloudy day



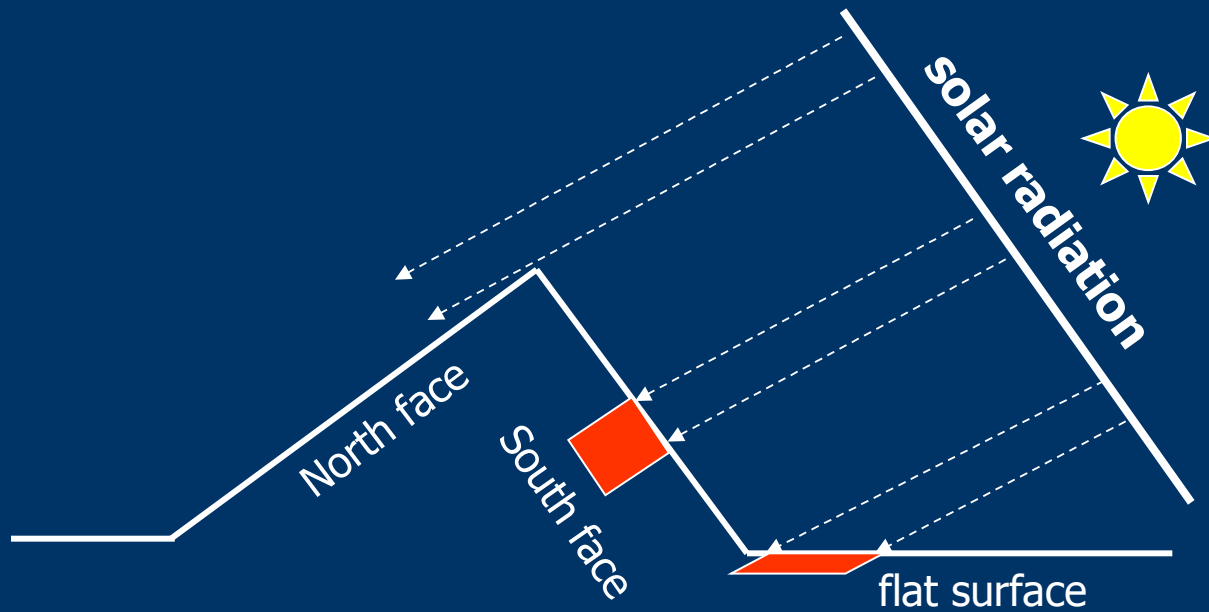
Incoming solar radiation at the earth's surface



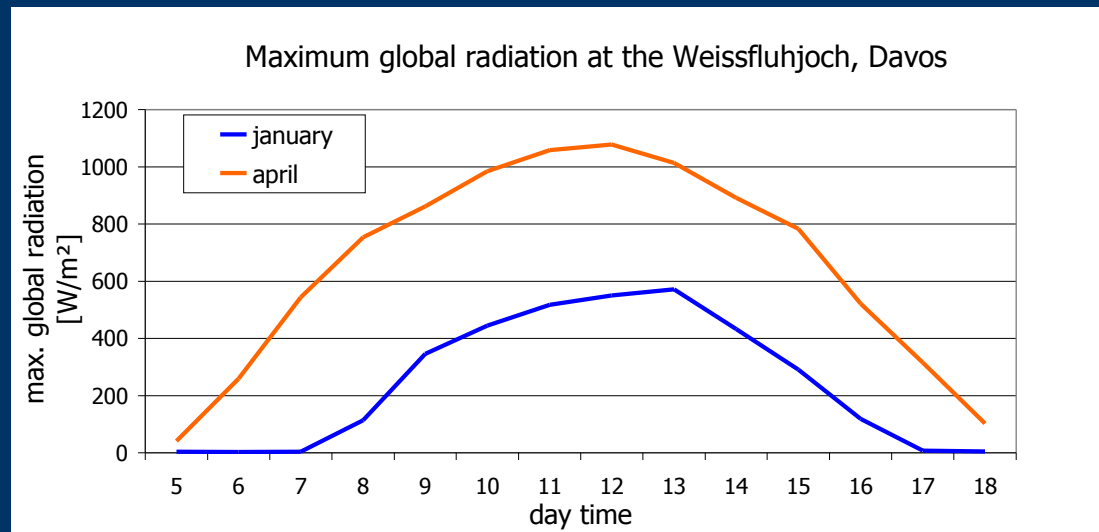
diffuse incoming solar radiation



Incoming solar radiation at the earth's surface



the energy input depends on the angle of incidence of the sun and on the season

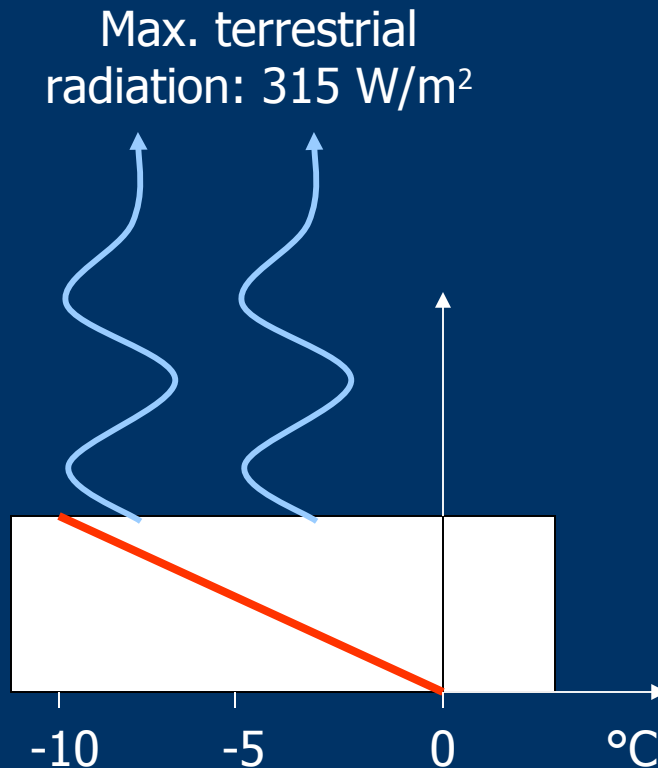


Terrestrial radiation

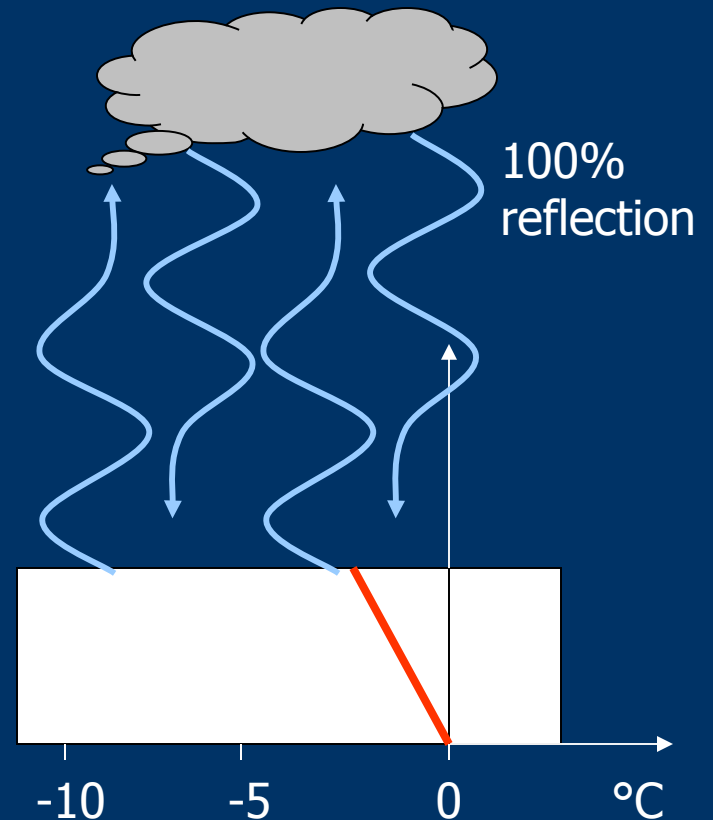


terrestrial radiation: day and night

clear day or night



cloudy day or night



Short and long wave radiation



solar radiation = short wave radiation ($\lambda = 0.5 \mu\text{m}$)

terrestrial radiation = long wave radiation ($\lambda = 10 \mu\text{m}$)

snow absorbs:

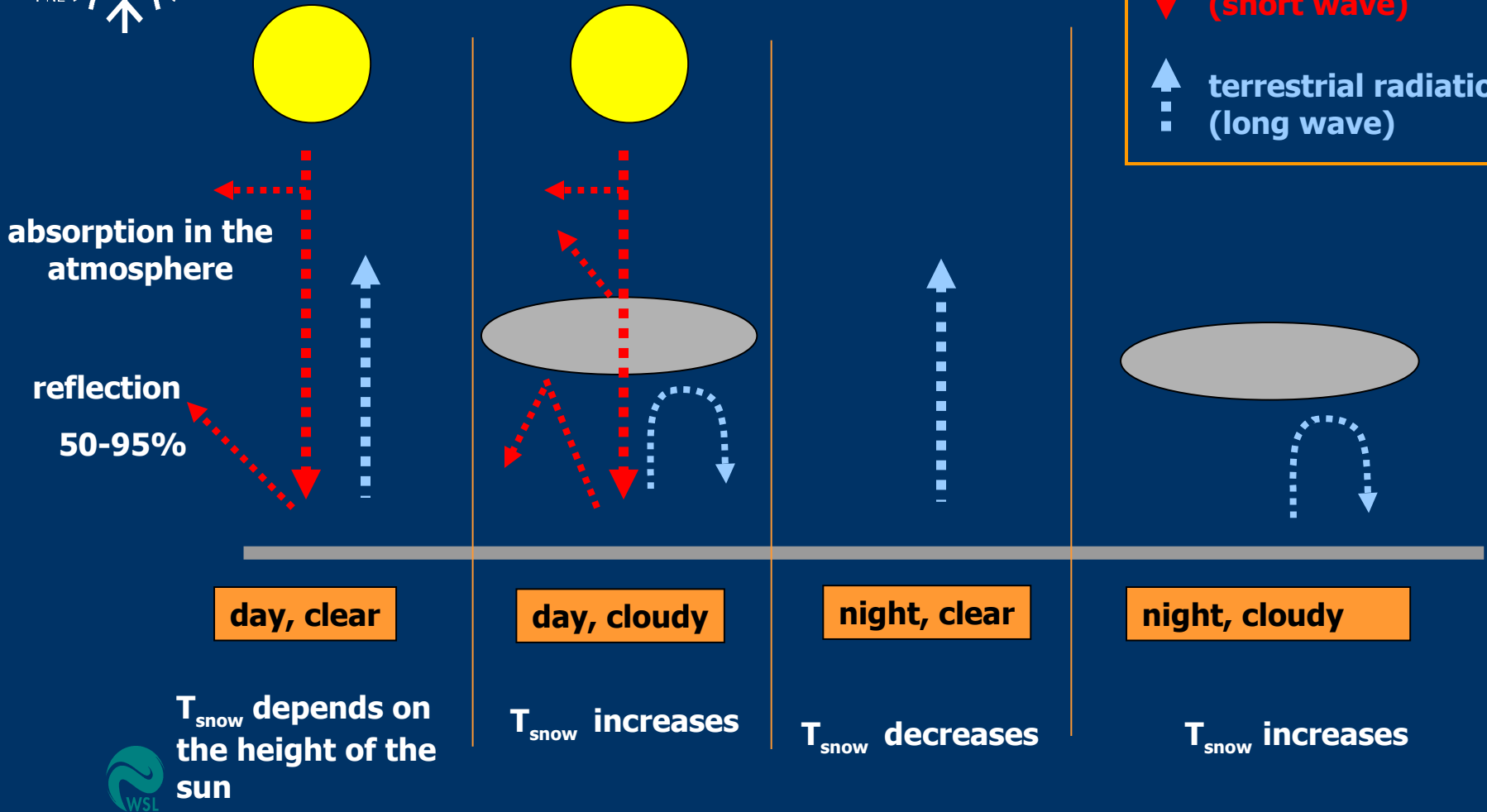
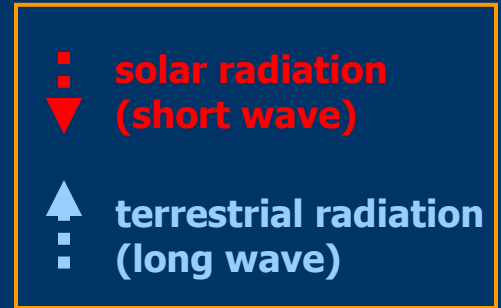
- 10 to 50 % of short wave radiation
- 99 % of long wave radiation

➡ long wave radiation has a bigger influence on the melting of snow than short wave radiation

➡ on a cloudy day, the snow's temperature increases faster due to the reflection of long wave radiation

Radiation balance at the snow surface

summary



Mechanical handling of snow



Mechanical handling of snow



What do we have ?

snow



What do we want ?

ski piste:

- **hard**

- **homogeneous**

{ bonds
density
temperature
L.W.C.

in order to obtain a good quality piste, we need to :

- prepare the snow mechanically
- optimise the preparation time
- use the natural snow transformation process

Mechanical handling of snow



combination of man and nature

mechanical handling is a
PREPARATION for the natural snow
solidification process

- diminution of mean grain size
- obtaining of different grain sizes
- grains become rounder
- increase of snow density

natural snow solidification

- settlement and sintering

snow hardening

man

nature



When does the snow have to be prepared?



The preparation time depends on:

- **type of snow** (temperature, liquid water content, grain shape)
- **weather forecasts**
(air temperature, air humidity, net radiation)



The preparation time must be chosen so that the snow hardening is the highest

When does the snow have to be prepared?



dry snow

- snow hardening by sintering needs time (settling time is crucial before using the piste)

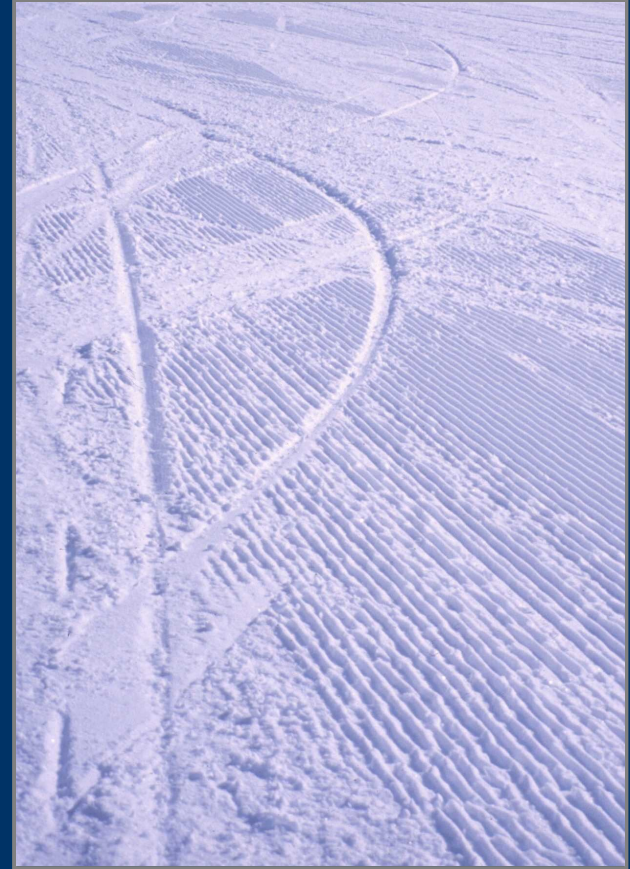
wet snow

- with a high liquid water content, the mechanical handling produces a water layer at the snow surface, which can freeze afterwards
- the best preparation time is just before the freezing starts

Influence of settling time on the sintering process



2 hours settling

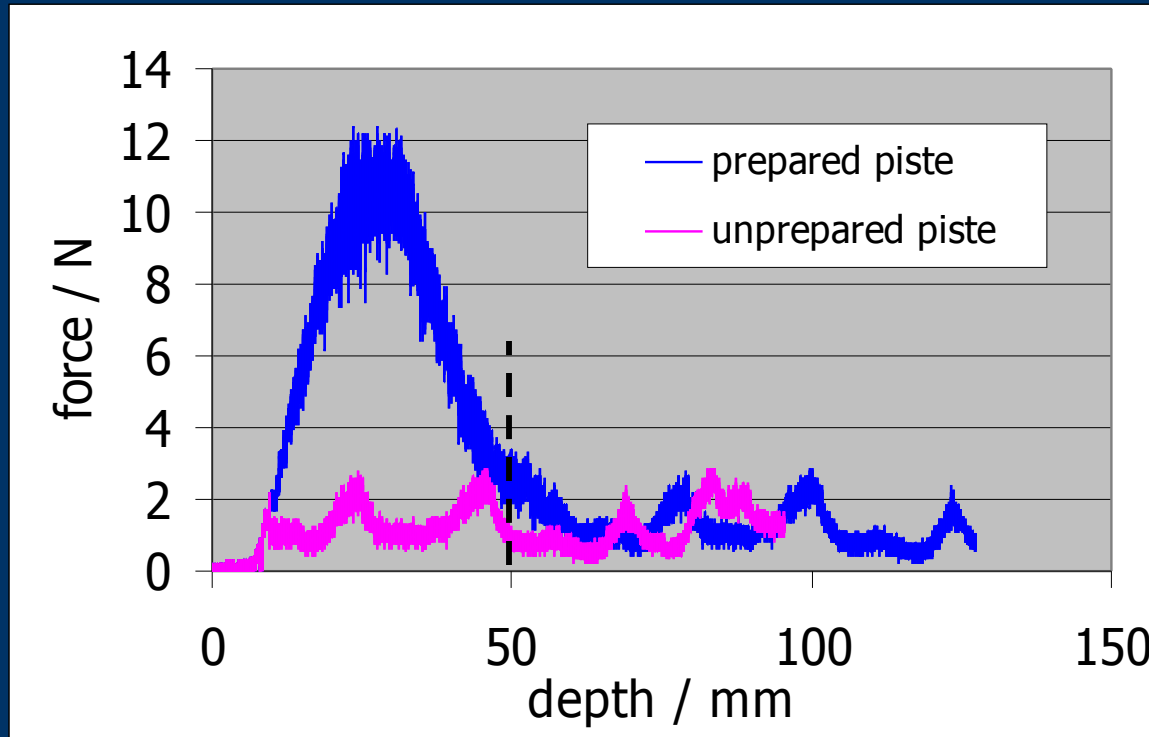


10 hours settling

Impact of grooming machines on the snow



example: preparation of new snow, $T_{\text{snow}} = -15^{\circ}\text{C}$



→ impact of the machine only at the surface (5 – 20 cm)



→ depends on: snow density, type of grain and snow temperature

Preparation and maintenance of race pistes



Race pistes must have high strengths in order to support the high forces produced by the skiers and the meteorological influences without transformation and therefore allowing a fair competition. → **hard piste**

Preparation and maintenance of race pistes



3 major tasks:

- building a hard fundament
- hardening the snow surface
- repairing the piste during the race

Building a hard fundament

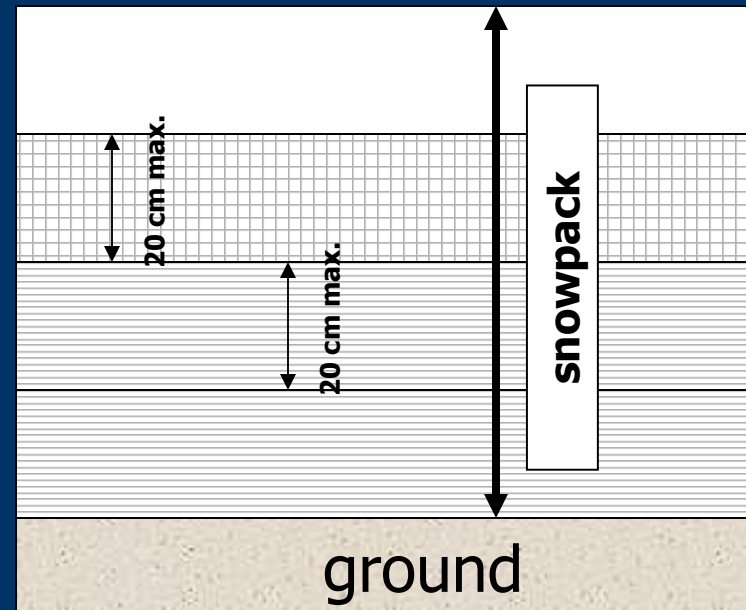


- since machine-made snow has high mechanical strength, it can constitute a good fundament
- preparation of new snow: compaction of successive snow layers (max. 20 cm thickness per layer)

preparation layer N°3
(machine, front blade and tiller) ➔

preparation layer N°2
(machine and front blade) ➔

preparation layer N°1
(machine and front blade) ➔



Hardening the snow surface



3 methods can be used to harden the snow surface:

- mechanical handling (grooming machines and skis)
- using of water
- using of chemicals



choice of the method depends on:

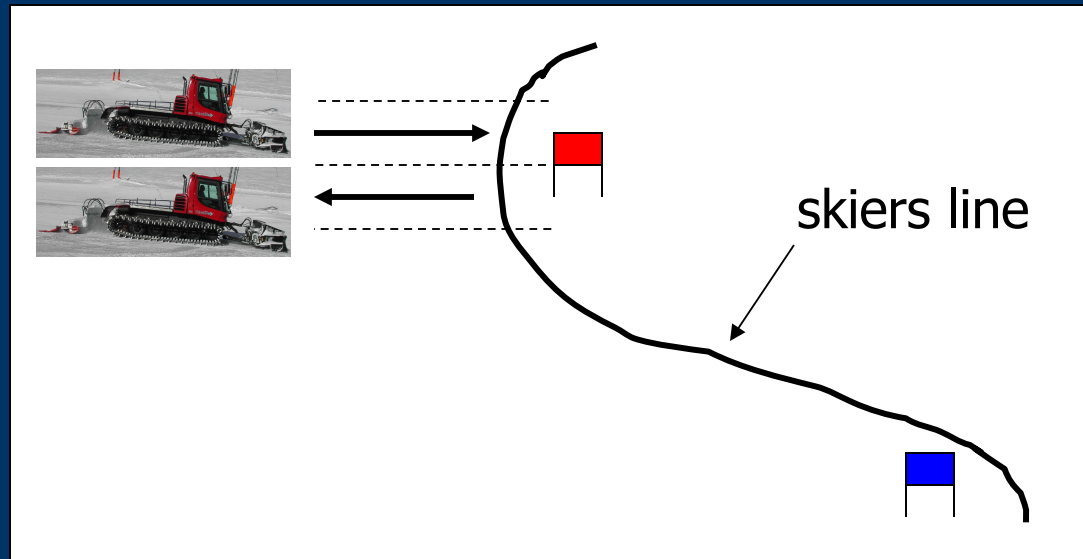
- snow type
- weather conditions
- race (downhill, super-G, slalom)

Hardening the snow surface



mechanical handling

- In the critical zones (curves, compressions, etc.) the piste must be, if possible, groomed perpendicular to the skiers line. The piste will therefore be more homogeneous and harder.



Hardening by water injection

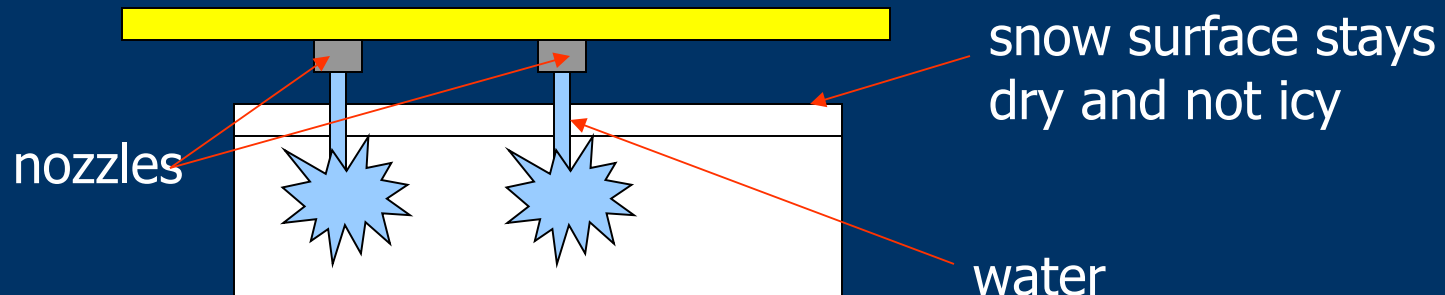


aim

- ➡ increase snow density and amount and size of bonds between snow grains

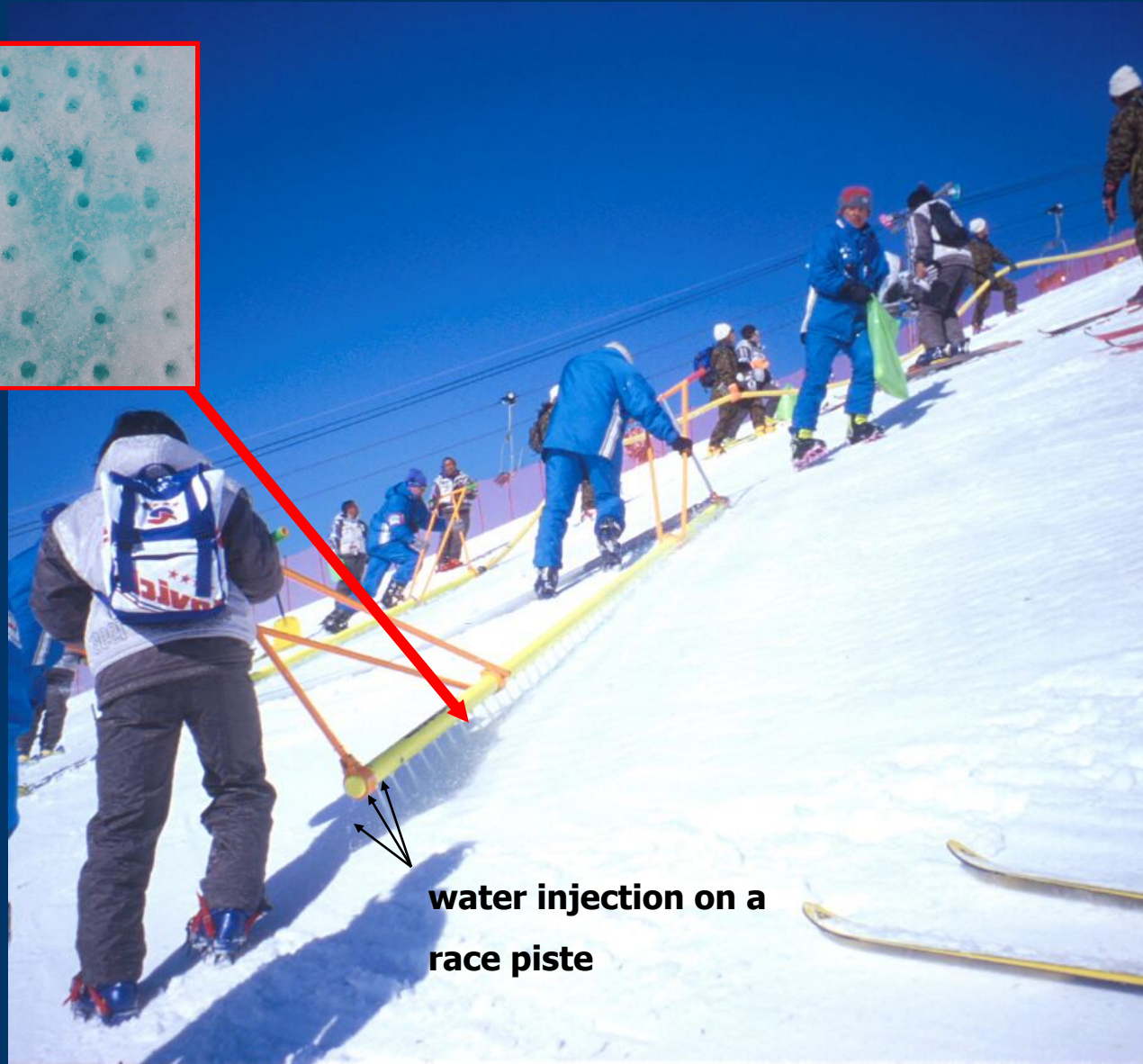
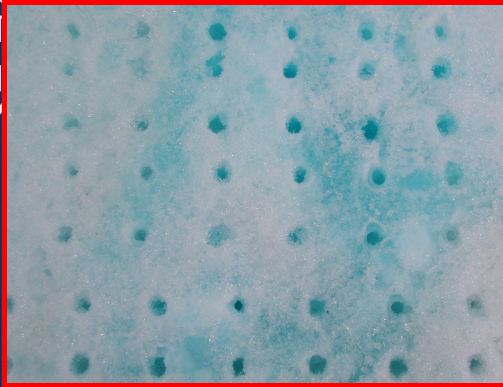
method

- ➡ injection of water with high pressure thru nozzles in the compacted snowpack
- ➡ Maximum:
 - 20 l/m² for fine grained snow ($\phi < 0.5$ mm)
 - 10 l/m² for coarse grained snow ($\phi > 1$ mm)
- ➡ pressure and flow can be modified



Hardening by water injection

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Hardening by water injection



physical process

- increase of snow density
- freezing of liquid water with $T_{\text{snow}} < 0^{\circ}\text{C}$
- heat is released
- heat is conducted out of the snowpack mostly by radiation and evaporation

conditions

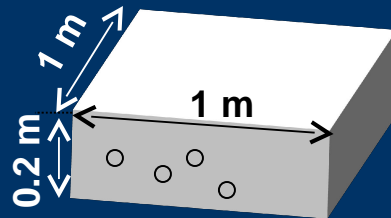
- ➡ snow temperature $< 0^{\circ}\text{C}$
- ➡ negative heat balance
- ➡ hard fundament

Hardening by water injection

Example: 1m x 1m x 0.2m

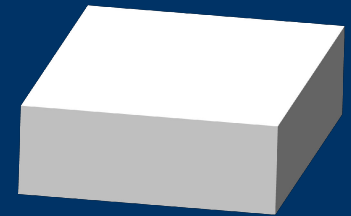
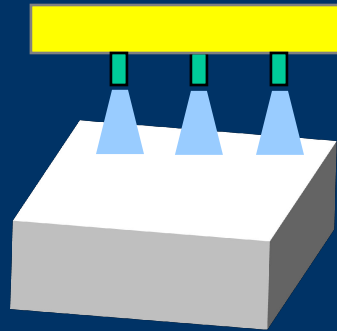


fine grained snow



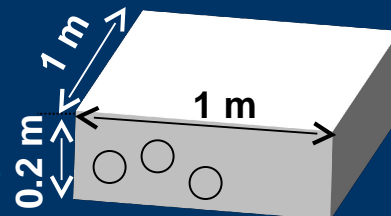
density = 300 kg/m³

**maximum water
injection: 20 l/m²**



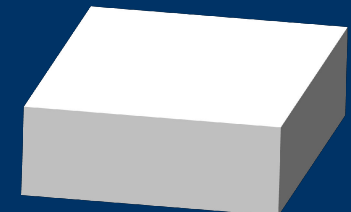
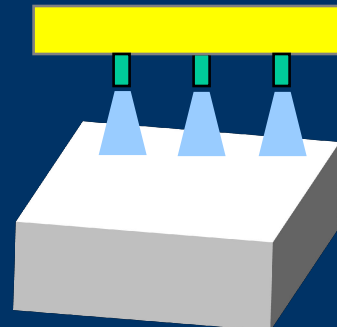
density = 400 kg/m³

coarse grained snow



density = 300 kg/m³

**maximum water
injection: 10 l/m²**



density = 350 kg/m³



Hardening by application of chemicals



physical process

- { dissolution of the chemicals in the liquid water
+ melting of an amount of snow
- needs heat
- temperature sinks
- freezing of the solution (water + chemicals)
- heat release that must be conducted out of the snowpack (mostly by radiation energy)

Hardening by application of chemicals



most widely used chemicals

- ➡ sodium chloride (cooking salt)
- ➡ ammonium nitrate
- ➡ combinations

method

- ➡ application of a defined amount (see manufacturer's guide) of chemicals on the snow surface
- ➡ side slipping with skis (by high solar radiation: instant slipping, by high terrestrial radiation: slipping after crystallisation)



The amount of chemicals has a big influence on the process.

Hardening by application of chemicals



conditions

- wet snow
- negative heat balance

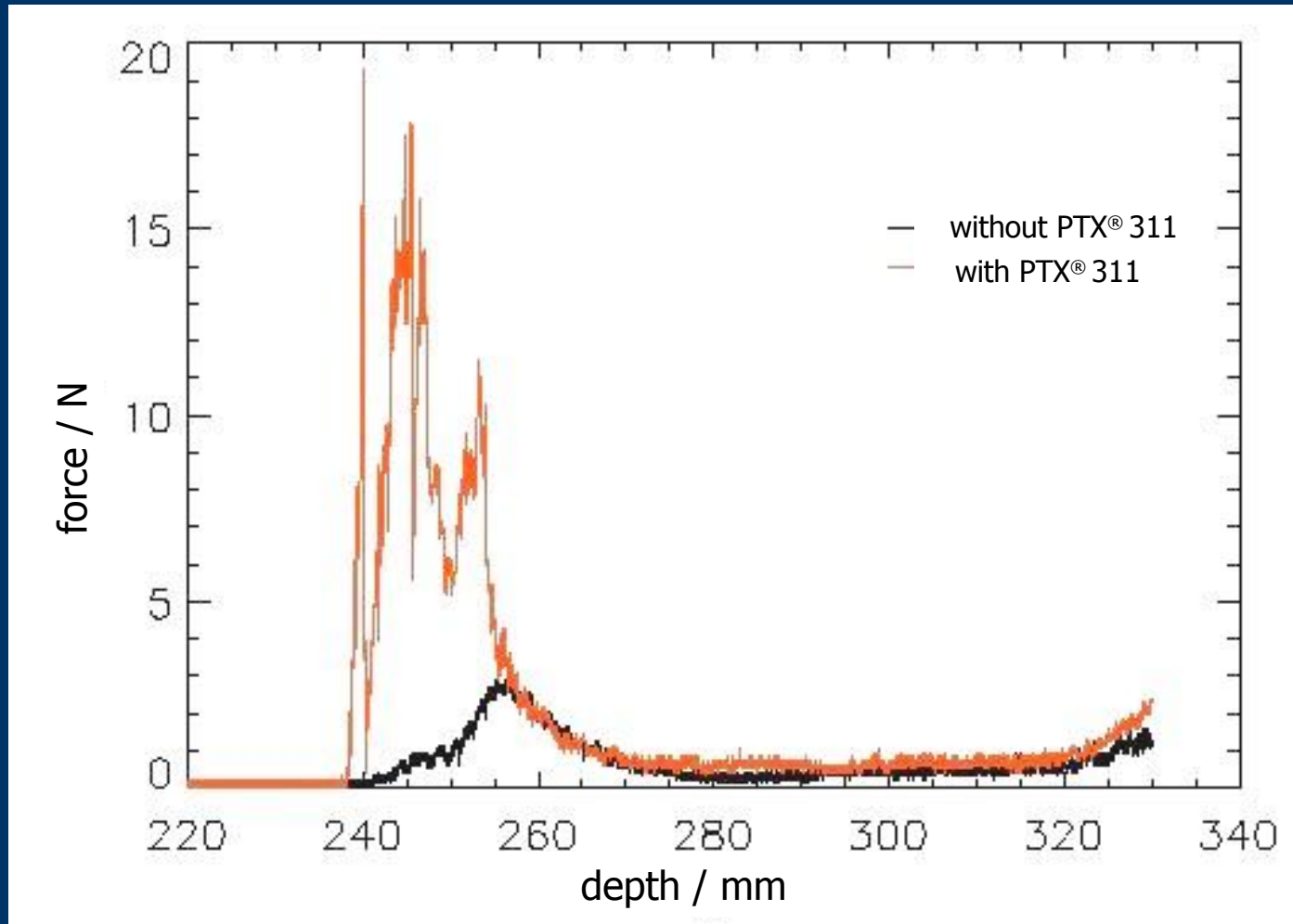


When snow is dry and no water can be sprayed to harden it, chemicals that release heat when they are mixed with snow, like calcium chloride, can be used to melt the snow.

Hardening by application of chemicals



example: application of PTX[®] 311 on a wet snow surface



Special race piste preparation methods



summary

dry new snow

density $< 400 \text{ kg/m}^3$

water (several
applications)

+

chemicals
(fasten the freezing
process)

dry snow

density $> 400 \text{ kg/m}^3$

water (event.
several applications)

wet snow

chemicals



Reparation of the piste during the race



possible damages

- holes
- ruts
- chatters

repairs

- level the piste
- fill in the holes
- spread the snow
- limit the damage



(lengthen the blended area)



shovels
rakes
skis

Practical example N°1

Preparation of a downhill race piste – new snow – cloudy night



snow conditions

- 20 cm new snow at the surface
- snow density = 150 kg/m^3
- $T_{\text{snow}} = -4^\circ\text{C}$
- hard fundament

weather forecasts

- cloudy night = constant snow temperature
- air temperature : between -1°C and -3°C

Practical example N°1

Preparation of a downhill race piste – new snow – cloudy night



preparation

glide zones:

- preparation with grooming machines directly after the snowfall
- settling time of at least 8 hours in order to obtain a sufficient natural snow consolidation

curves/jumps/compressions:

- remove the snow out of the track with shovels or eventually snow tillers
- warning: snow on fall-zones (side of the track) must also be prepared and the nets must be snow-free



Practical example N°2

Preparation of a slalom piste – dry and weak snow



snow conditions

- snow surface density = 450 kg/m^3
- fine grained snow ($\phi < 0.5 \text{ mm}$)
- $T_{\text{snow}} = -8^\circ\text{C}$

weather forecasts

- clear day (high terrestrial radiation)

Practical example N°2

Preparation of a slalom piste – dry and weak snow



preparation

- water injection No. 1 with max. 20 l/m²
(density: 450 -> 550 kg/m³)
- freezing time (minimum 6 hours)
- water injection No. 2 with max. 10 l/m²
(density: 550 -> 600 kg/m³)
- freezing time (minimum 4 hours)

Practical example N°3

Preparation of a super-G piste – wet and weak snow – competition day



snow conditions

- snow surface density = 550 kg/m^3
- coarse grained snow
- wet, liquid water content = 7%, $T_{\text{snow}} = 0^\circ\text{C}$

weather conditions

- clear day (terrestrial radiation)
- high solar radiation (spring)

preparation

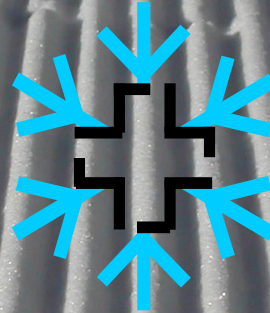
- application of PTX® 311
- packing and slipping with skis



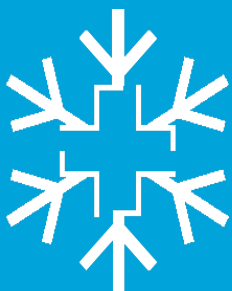
Thank you for your attention



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Swiss Federal Institute for Snow and Avalanche Research SLF



Preparation of alpine ski pistes

Preparation of alpine ski pistes



This Power Point Presentation is intended for FIS alpine technical delegated persons. It has been written by Mathieu Fauve and Hansueli Rhyner from the Swiss Federal Institute for Snow and Avalanche Research, Davos and completed by Thomas Gurzeler. It is a tool for the alpine race supervisors, officials, coaches and Athletes, who strive for a safe race. The use and modifications of this presentation are possible only with permission from Hansueli Rhyner.

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Preparation of alpine ski pistes



Aim

The aim is to transform a soft snow into a:

- hard
- homogeneous

piste

Proceed

Use of **mechanical equipment** taking into account the **physical properties of snow** and the **meteorology**

Special

Race pistes have to be prepared with **special methods** in order to obtain very high strength



Preparation of alpine skiing slopes



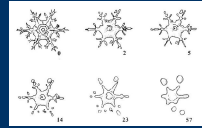
physical properties of snow

snow and meteorology

mechanical handling of snow



preparation and maintenance of race
pistes



Physical properties of snow



Snow is a very particular material:

- ➔ composed of air and water in all its forms: solid (ice), gas (water vapour) and possibly liquid (liquid water)
- ➔ near its melting point (0°C), so extremely sensitive to variations of temperature and pressure and reacts rapidly
- ➔ exists in different forms and is subjected to a continuous transformation



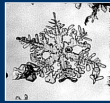
Comparison: If we bring the temperature of a metal, e.g. aluminium, to 20°C below its melting point, it becomes unstable and reacts strongly to influences of the environment.

A snow crystal transforms itself continuously from its formation until its melting.

Snow grains

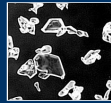


new snow



+

faceted crystals



□

fragmented precipitation particles



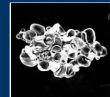
/

surface hoar



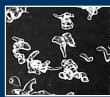
V

melt-freeze snow



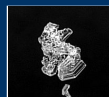
○

rounded grains



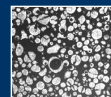
●

depth hoar



^

machine-made snow

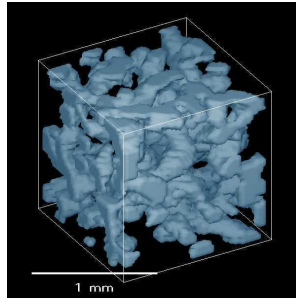


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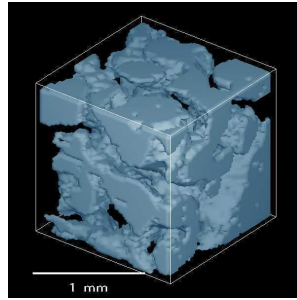


Depth hoar does not built up on ski pistes. Dense snow does not allow grain size to increase considerably.

Snow density



low density snow (220 kg/m³)

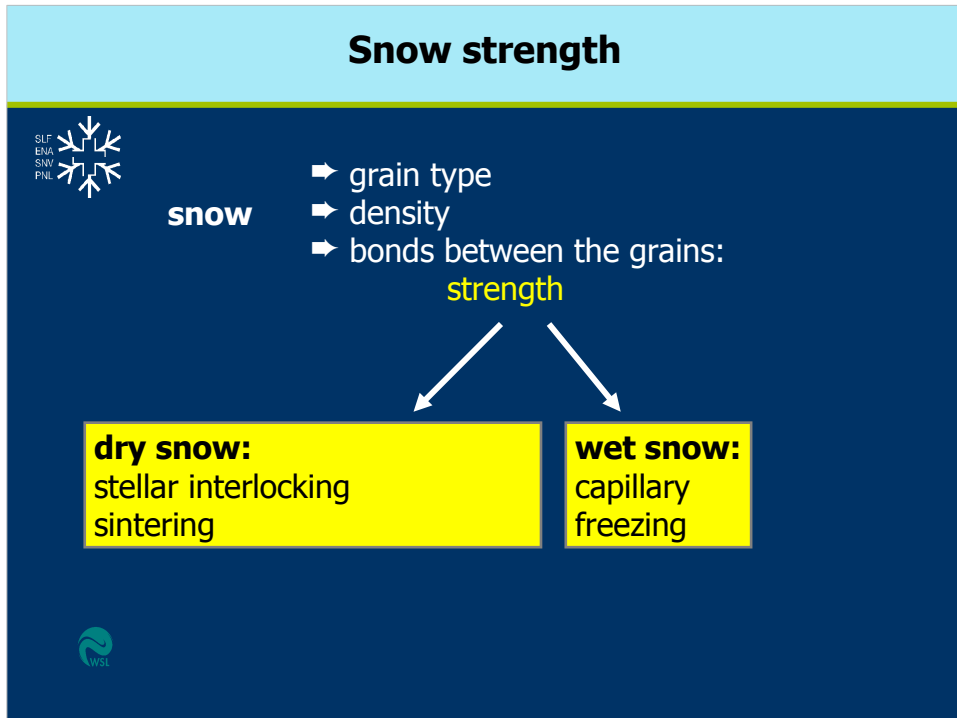


high density snow (516 kg/m³)

- new snow density: between 50 and 250 kg/m³
- average ski slope density : 480 kg/m³
- downhill racing slope density: 300 - 500 kg/m³
- super-G racing slope density: approx. 550 kg/m³
- slalom racing slope density: approx. 600 kg/m³



Warning: high density does not always mean high strength



The bonds between snow grains are the most important parameter responsible for the resistance of snow

Strength by stellar interlocking



- concerns only new snow stellar crystals
- branches are connected to each other



→ weak and non-lasting strength

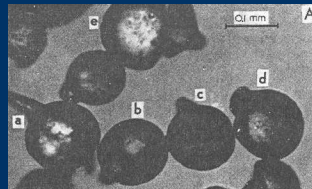


Not relevant on ski slopes

Strength by sintering



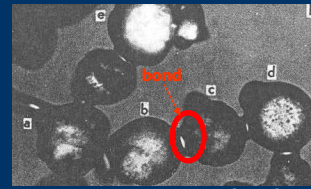
sintering: building of solid bonds between snow grains by water vapour transport



$T = -1.5^{\circ}\text{C}$



after 165 min



(Kuroiwa, 1974)

► the building of the bonds and the increasing of their size depend on the following parameters:

- **temperature** (process faster near 0°C)
- **grain type** (shape, mean size, size distribution)
- **density** (number of contact points)
- **time** (process needs time)



Grain shape: as round as possible

Grain size distribution: different grain sizes in order to fill as many pores as possible

Strength of wet snow



- liquid water is held on grains by capillarity as long as its volume is lower than 5 to 10% (depending on the grain size) of the total volume. Small grains can hold more water than bigger ones.
- when the liquid water content (L.W.C.) is high, bonds melt and snow becomes softer
- when the liquid water freezes, strong bonds built between the grains



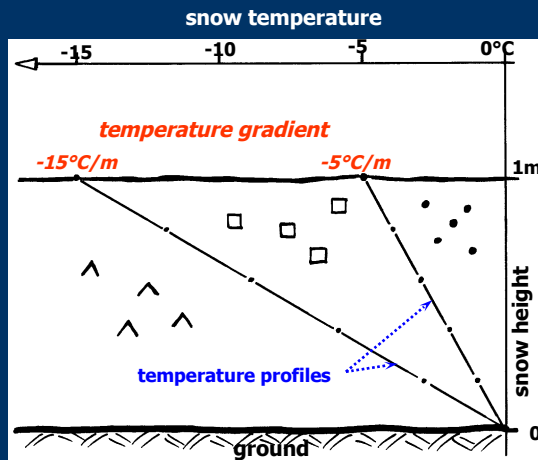
Most of the equipments used in the field do not measure the liquid water content.

In order to determine the liquid water content of snow by measuring its dielectric constant, we must know its density.

Snow metamorphism



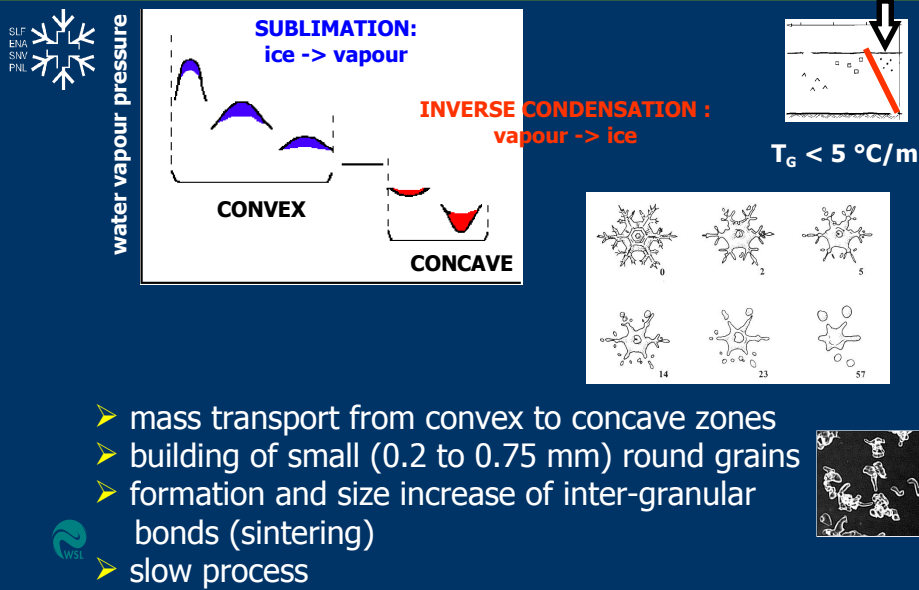
Snow metamorphism depends on the temperature gradient in the snowpack



$$T_G = \frac{dT}{dz}$$

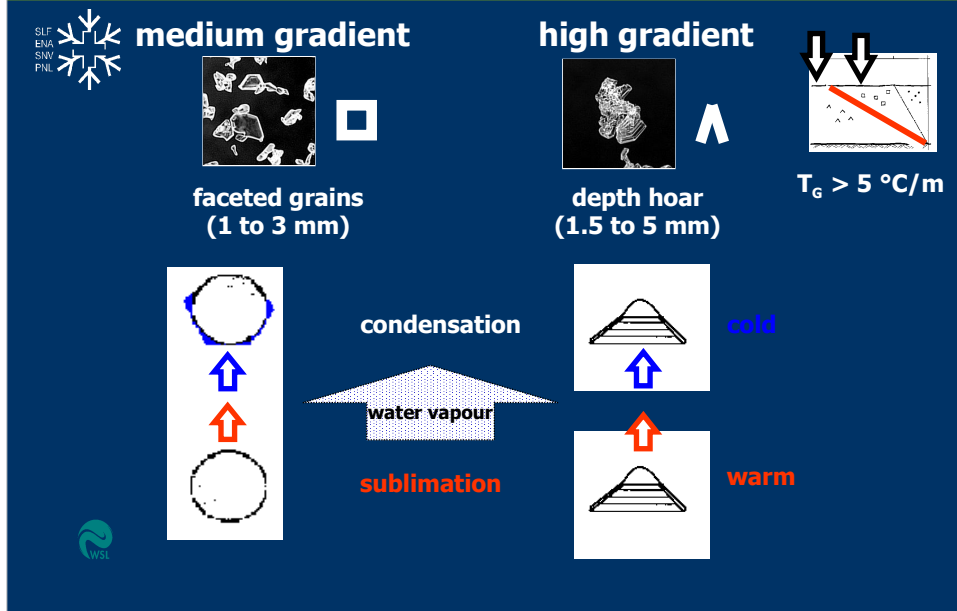


Destructive metamorphism



Destructive metamorphism leads to the settlement of snow and to its solidification

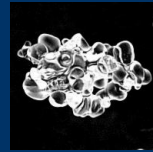
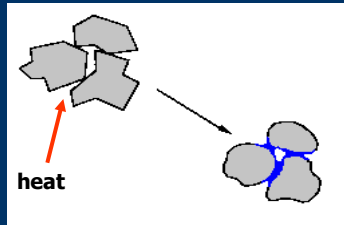
Constructive metamorphism



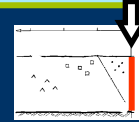
Constructive metamorphism = increase of grain size and loss of resistance

No formation of depth hoar on ski pistes (too high density)

Melt-freeze metamorphism



0.5 to 4 mm



$T = 0^{\circ}\text{C}$
 $\text{L.W.C.} > 0$

- grains become rounder and bigger
- influence of the liquid water content:
 - non-saturated snow (L.W.C. < 8 to 15% vol.)
 - ➔ clusters
 - saturated snow (L.W.C. > 8 to 15% vol.)
 - ➔ no cohesion



Snow can contain liquid water only when its temperature is equal 0°C

Mechanical properties of snow



The mechanical properties of snow mostly depend on the following parameters:

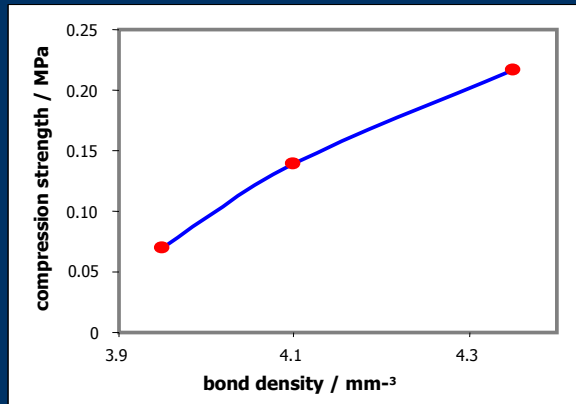
- bonds
- density
- temperature
- liquid water content



Mechanical properties of snow



bonds

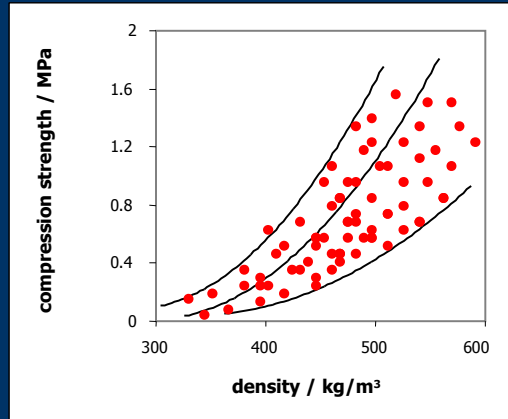


➡ the more bonds that exist and the larger they are, the higher the snow's resistance is

Mechanical properties of snow



density



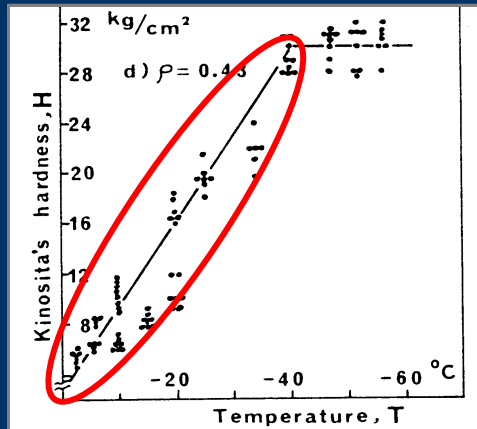
➡ usually, the denser the snow, the more resistant and tough it is

Example: wet spring snow has high density but low strength

Mechanical properties of snow



temperature



(Tusima, 1974)

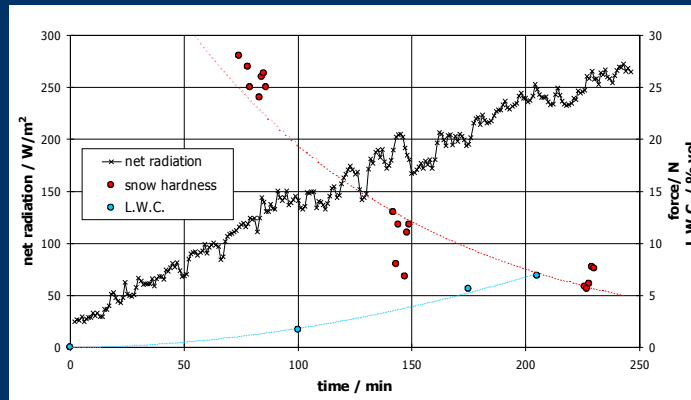


- the colder dry snow is, the more resistant and tough it becomes

Mechanical properties of snow



liquid water content ($T_{\text{snow}}=0^{\circ}\text{C}$)



➡ for a snow at 0°C and high solar incoming radiation, the liquid water content at the snow surface increases rapidly and snow becomes softer



machine-made snow

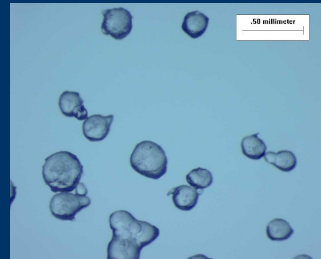
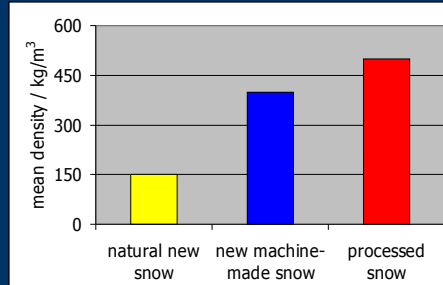


characteristics

- high density:
300 to 500 kg/m³
- small and round grains
(0.1 to 0.9 mm)



- ➡ resistant snow
- ➡ needs little compaction

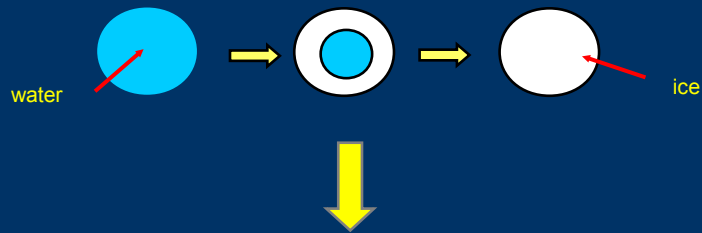


machine-made snow



characteristics

- Warning: risks of incomplete freezing of water droplets



➡ a curing time (for complete freezing) is needed before grooming the snow



In comparison with natural snow, machine-made snow freezes from the outside to the inside

Snow and meteorology



The different properties of snow depend mostly on one parameter: its temperature.
Snow temperature depends on heat exchanges between snow and air: heat balance

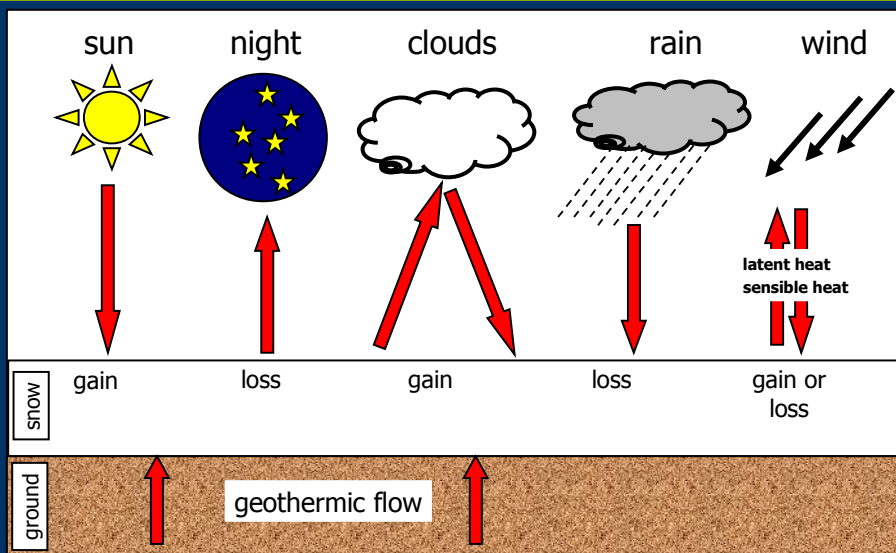
The heat balance at the snow surface = difference between gain and loss of heat energy



balance is positive
snow temperature increases
at 0°C: snow starts melting

balance is negative
snow temperature
decreases

Heat balance

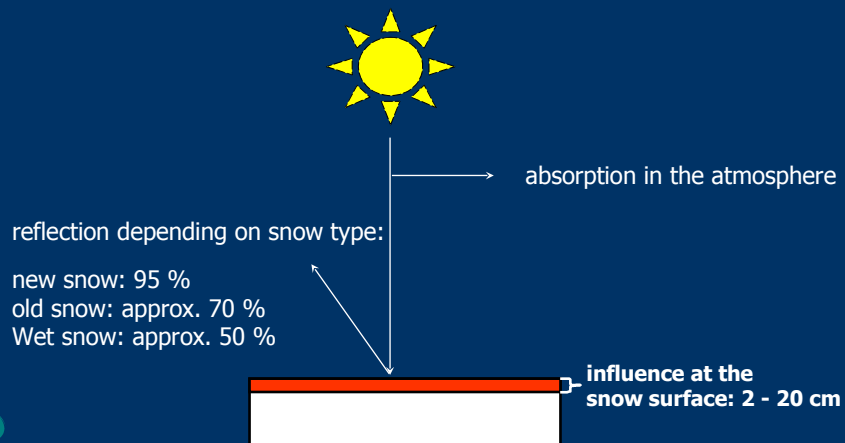


Gain and loss of heat energy at the snow surface

Incoming solar radiation at the earth's surface



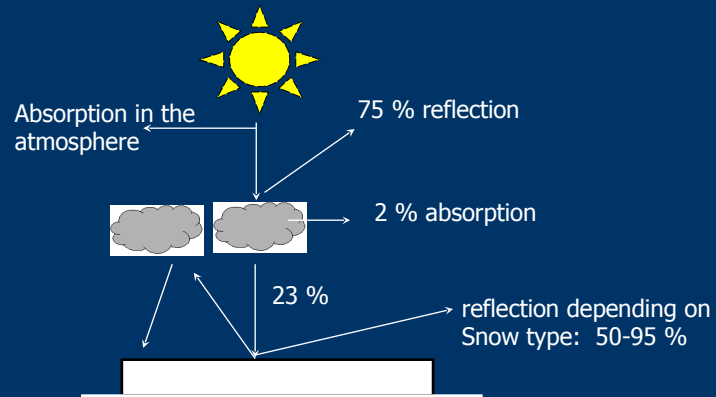
incoming solar radiation at the snow surface on a sunny day



Incoming solar radiation at the earth's surface



incoming solar radiation at the snow surface on a cloudy day

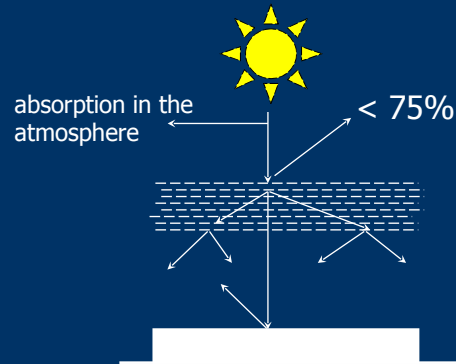


Involves an increase of snow temperature on all slope orientations

Incoming solar radiation at the earth's surface

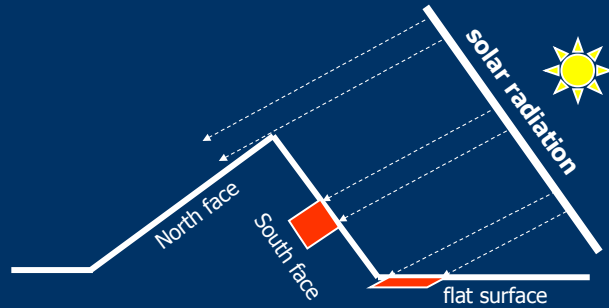


diffuse incoming solar radiation

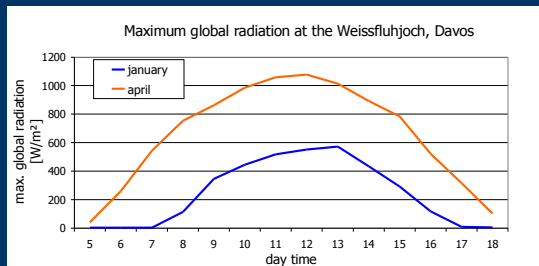


Involves an increase of snow temperature on all slope orientations

Incoming solar radiation at the earth's surface



the energy input depends on the angle of incidence of the sun and on the season



Example: World championship in Crans Montana 1986, February, Women Downhill: Start slope = South face, angle of incidence: 90° : high solar radiation: melting of snow (wet snow) / Flat zone in the forest: snow temperature = -12°C (low solar radiation)

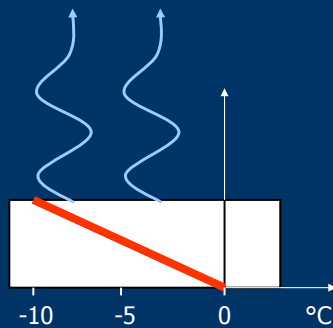
Terrestrial radiation



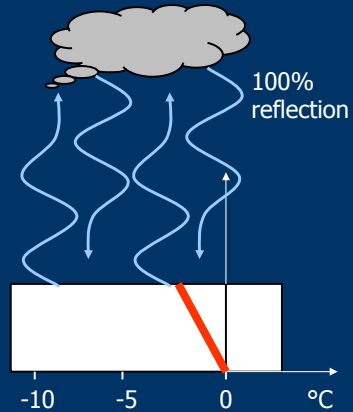
terrestrial radiation: day and night

clear day or night

Max. terrestrial
radiation: 315 W/m^2



cloudy day or night



Each body on the earth's surface emits radiation

Short and long wave radiation



solar radiation = short wave radiation ($\lambda = 0.5 \mu\text{m}$)

terrestrial radiation = long wave radiation ($\lambda = 10 \mu\text{m}$)

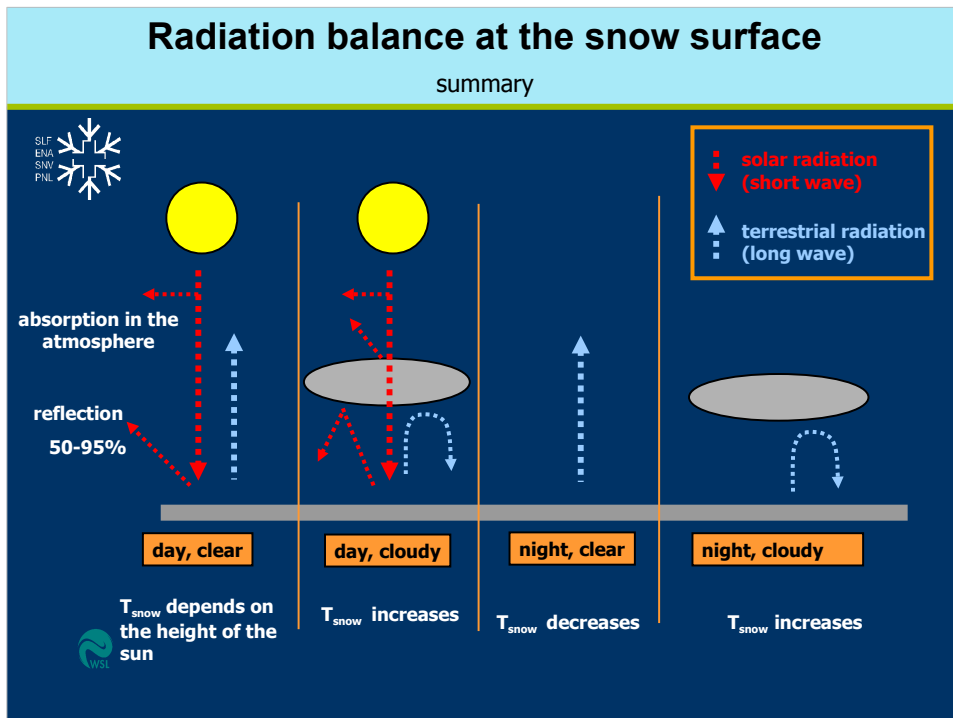
snow absorbs:

- 10 to 50 % of short wave radiation
- 99 % of long wave radiation

➡ long wave radiation has a bigger influence on the melting of snow than short wave radiation

➡ on a cloudy day, the snow's temperature increases faster due to the reflection of long wave radiation





Clear day: on locations where solar radiation is low (north face in winter), terrestrial radiation is often higher than solar radiation

Mechanical handling of snow



Mechanical handling of snow



What do we have ?

snow ✓

What do we want ?

ski piste:

- **hard**

- **homogeneous**

{ bonds
density
temperature
L.W.C.

in order to obtain a good quality piste, we need to :

- prepare the snow mechanically
- optimise the preparation time
- use the natural snow transformation process



Mechanical handling of snow



combination of man and nature

mechanical handling is a
PREPARATION for the natural snow
solidification process

- diminution of mean grain size
- obtaining of different grain sizes
- grains become rounder
- increase of snow density

natural snow solidification

- settlement and sintering

snow hardening



man

nature

When does the snow have to be prepared?



The preparation time depends on:

- **type of snow** (temperature, liquid water content, grain shape)
- **weather forecasts** (air temperature, air humidity, net radiation)



The preparation time must be chosen so that the snow hardening is the highest



When does the snow have to be prepared?



dry snow

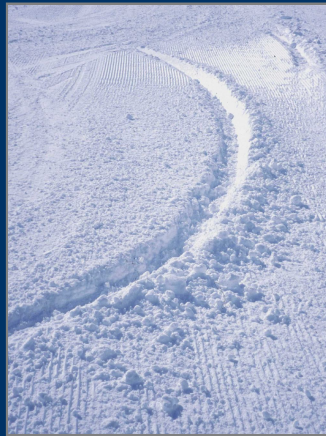
- snow hardening by sintering needs time (settling time is crucial before using the piste)

wet snow

- with a high liquid water content, the mechanical handling produces a water layer at the snow surface, which can freeze afterwards
- the best preparation time is just before the freezing starts



Influence of settling time on the sintering process



2 hours settling



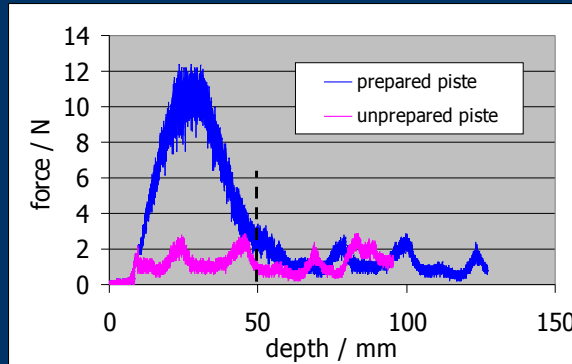
10 hours settling



Impact of grooming machines on the snow



example: preparation of new snow, $T_{\text{snow}} = -15^{\circ}\text{C}$



→ impact of the machine only at the surface (5 – 20 cm)



→ depends on: snow density, type of grain and snow temperature

Preparation and maintenance of race pistes



(photo: Stöckli)

Race pistes must have high strengths in order to support the high forces produced by the skiers and the meteorological influences without transformation and therefore allowing a fair competition. → hard piste



Preparation and maintenance of race pistes



3 major tasks:

- building a hard fundament
- hardening the snow surface
- repairing the piste during the race



Building a hard fundement

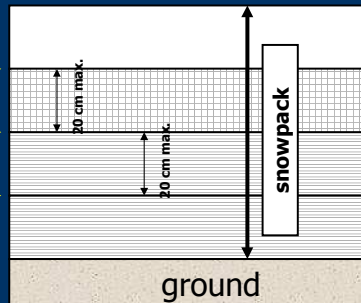


- since machine-made snow has high mechanical strength, it can constitute a good fundement
- preparation of new snow: compaction of successive snow layers (max. 20 cm thickness per layer)

preparation layer N°3
(machine, front blade and tiller)

preparation layer N°2
(machine and front blade)

preparation layer N°1
(machine and front blade)



Hardening the snow surface



3 methods can be used to harden the snow surface:

- mechanical handling (grooming machines and skis)
- using of water
- using of chemicals



choice of the method depends on:

- snow type
- weather conditions
- race (downhill, super-G, slalom)

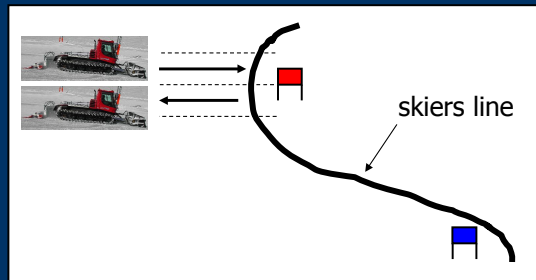


Hardening the snow surface



mechanical handling

- In the critical zones (curves, compressions, etc.) the piste must be, if possible, groomed perpendicular to the skiers line. The piste will therefore be more homogeneous and harder.



The picture is exaggerated. The preparation must take account of the topography and the machines possibilities.

Important is to prepare the snow as perpendicular as possible to the skiers line.

Hardening by water injection

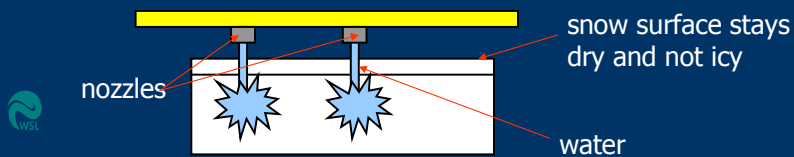


aim

- ➔ increase snow density and amount and size of bonds between snow grains

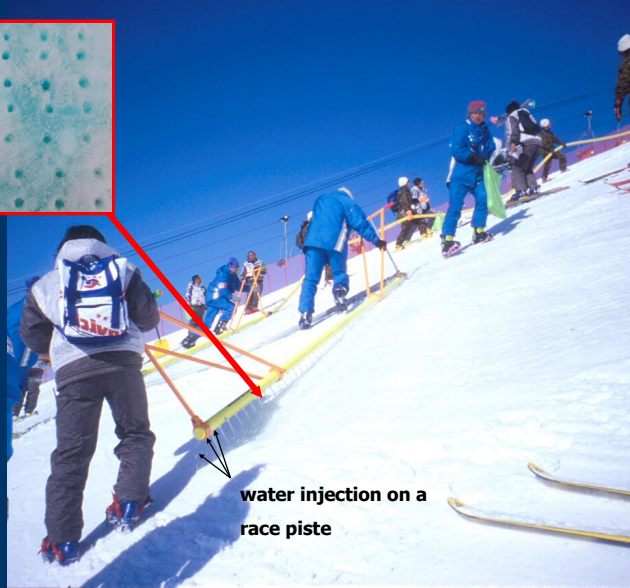
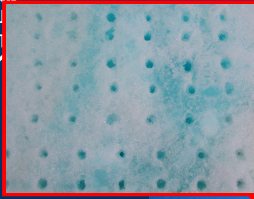
method

- ➔ injection of water with high pressure thru nozzles in the compacted snowpack
- ➔ Maximum:
 - 20 l/m² for fine grained snow ($\phi < 0.5$ mm)
 - 10 l/m² for coarse grained snow ($\phi > 1$ mm)
- ➔ pressure and flow can be modified



Hardening by water injection

SILF
ENR
SIVR
PNL



water injection on a
race piste



Hardening by water injection



physical process



increase of snow density

freezing of liquid water with $T_{\text{snow}} < 0^{\circ}\text{C}$

heat is released

heat is conducted out of the snowpack mostly by radiation and evaporation

conditions

- snow temperature $< 0^{\circ}\text{C}$
- negative heat balance
- hard fundament

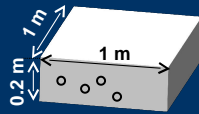


Hardening by water injection

Example: 1m x 1m x 0.2m

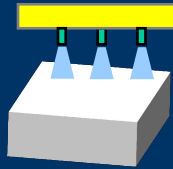


fine grained snow



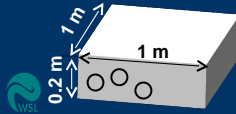
density = 300 kg/m³

maximum water
injection: 20 l/m²



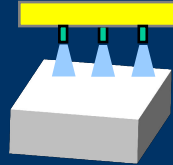
density = 400 kg/m³

coarse grained snow



density = 300 kg/m³

maximum water
injection: 10 l/m²



density = 350 kg/m³

Hardening by application of chemicals



physical process

- { dissolution of the chemicals in the liquid water
- + melting of an amount of snow
- needs heat
- temperature sinks
- freezing of the solution (water + chemicals)
- heat release that must be conducted out of the snowpack (mostly by radiation energy)



Hardening by application of chemicals



most widely used chemicals

- ➡ sodium chloride (cooking salt)
- ➡ ammonium nitrate
- ➡ combinations

method

- ➡ application of a defined amount (see manufacturer's guide) of chemicals on the snow surface
- ➡ side slipping with skis (by high solar radiation: instant slipping, by high terrestrial radiation: slipping after crystallisation)



The amount of chemicals has a big influence on the process.

Hardening by application of chemicals



conditions

- wet snow
- negative heat balance



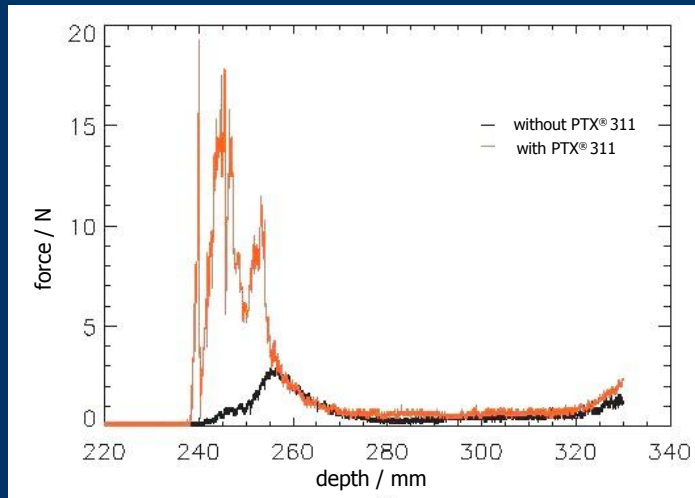
When snow is dry and no water can be sprayed to harden it, chemicals that release heat when they are mixed with snow, like calcium chloride, can be used to melt the snow.



Hardening by application of chemicals



example: application of PTX® 311 on a wet snow surface



Measurements in St-Moritz on a clear day

Special race piste preparation methods



summary

dry new snow density < 400 kg/m ³	dry snow density > 400 kg/m ³	wet snow
water (several applications) + chemicals (fasten the freezing process)	water (event. several applications)	chemicals



Reparation of the piste during the race



possible damages

- holes
- ruts
- chatters

reparations

- level the piste
- fill in the holes
- spread the snow
- limit the damage

(lengthen the blended area)



} shovels
rakes
skis



Practical example N°1

Preparation of a downhill race piste – new snow – cloudy night



snow conditions

- 20 cm new snow at the surface
- snow density = 150 kg/m^3
- $T_{\text{snow}} = -4^\circ\text{C}$
- hard fundament

weather forecasts

- cloudy night = constant snow temperature
- air temperature : between -1°C and -3°C



Practical example N°1

Preparation of a downhill race piste – new snow – cloudy night



preparation

glide zones:

- preparation with grooming machines directly after the snowfall
- settling time of at least 8 hours in order to obtain a sufficient natural snow consolidation

curves/jumps/compressions:

- remove the snow out of the track with shovels or eventually snow tillers
- warning: snow on fall-zones (side of the track) must also be prepared and the nets must be snow-free



Practical example N°2

Preparation of a slalom piste – dry and weak snow



snow conditions

- snow surface density = 450 kg/m^3
- fine grained snow ($\phi < 0.5 \text{ mm}$)
- $T_{\text{snow}} = -8^\circ\text{C}$

weather forecasts

- clear day (high terrestrial radiation)



Practical example N°2

Preparation of a slalom piste – dry and weak snow



preparation

- water injection No. 1 with max. 20 l/m²
(density: 450 -> 550 kg/m³)
- freezing time (minimum 6 hours)
- water injection No. 2 with max. 10 l/m²
(density: 550 -> 600 kg/m³)
- freezing time (minimum 4 hours)



Practical example N°3

Preparation of a super-G piste – wet and weak snow – competition day



snow conditions

- snow surface density = 550 kg/m^3
- coarse grained snow
- wet, liquid water content = 7%, $T_{\text{snow}} = 0^\circ\text{C}$

weather conditions

- clear day (terrestrial radiation)
- high solar radiation (spring)

preparation



- application of PTX[®] 311
- packing and slipping with skis

Thank you for your attention

